



# Geochemistry of apatite individuals in Zhijin phosphorites, South China: Insight into the REY sources and diagenetic enrichment

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## ABSTRACT

Marine sedimentary phosphorites are rich in rare earth elements (REE) and yttrium (REY) which primarily hosted in colophonite composed of closely-packed apatite nanocrystals. Limited by small particle diameters, the fine metallogenic process of apatite crystals and REY enrichment remains unknown. To address this, using early Cambrian REY-rich phosphorite from Zhijin, South China as an example, mineralogical and in situ geochemical characteristics of apatite individuals were analyzed and compared with colophonites. Our study found that: (1) the apatite individuals, with diameters between 40 and 100  $\mu\text{m}$ , were divided into authigenic/diagenetic apatites, terrigenous clastic apatites, and hydrothermally altered apatites; (2) Terrigenous clastic apatites have highest  $\sum\text{REY}$  levels (4673–12232 ppm) and lowest Y/Ho (23–28), in addition, Y/Ho showed negative relationships with  $\sum\text{REY}$  in all apatite monominerals, as well as bulk phosphorites; (3) Compared with colophonites and old phosphorite, authigenic/diagenetic apatite individuals showed similar REY patterns, in addition, diagenetic apatites have higher  $\sum\text{REY}$ , Ce/Ce\* and Eu/Eu\* ratios but lower  $(\text{La}/\text{Yb})_N$  and  $(\text{La}/\text{Sm})_N$  ( $>0.5$ ) ratios relative to authigenic apatite individuals and colophonites. This dataset indicated that weathering of terrestrial magmatic rocks provided not only clastic apatites but also REY into the South China ocean during the early Cambrian, mixed with REY derived from deep seawater. Authigenic apatite individuals form similarly to apatite nanocrystals in colophonites, with diagenetic modification playing vital role. During this process, apatite individuals grew up and underwent stronger diagenesis than apatite nanocrystals in colophonites, absorbing REY from porewater, which resulted in higher REY concentrations in apatite individuals. This study provides new data and insight into the formation of authigenic/diagenetic apatite individuals and improves our understanding of REY sources and diagenetic effects on REY enrichment.

## 1. Introduction

Rare earth elements plus Y (REY) are important resources in 21st century, widely used in new technology industries, which include national defense, military, aerospace, clean energy, special materials, and others. Recently, high-tech industries increased the global demand for REY. Most countries have listed REY as a key mineral resource and attached great importance to studying REY mineralization. Typical REY deposits come from carbonate rocks, alkaline rocks/alkaline granite, such as Bayan Obo, Yinachang, and Mianning-Dechang deposits (Ye

et al., 2013; Li and Zhou, 2015; Fan H.R. et al., 2016; Liu et al., 2019). Furthermore, supergenetic weathering type/ion adsorption clay-type (Bao and Zhao, 2008; Wang et al., 2018), placer type (Yuan et al., 2012), and sedimentary type (deep-sea mud and phosphorite) (Kato et al., 2011; Emsbo et al., 2015) could provide REY resources. Among them, phosphorite contained enriched levels of middle REEs (MREE) and especially scarce heavy REEs (HREEs) (Emsbo et al., 2015).

Sedimentary phosphorite deposits occur worldwide, especially in Jordan, North America, South China, Australia, and South Africa. Furthermore, phosphorite formed across multiple geological periods,

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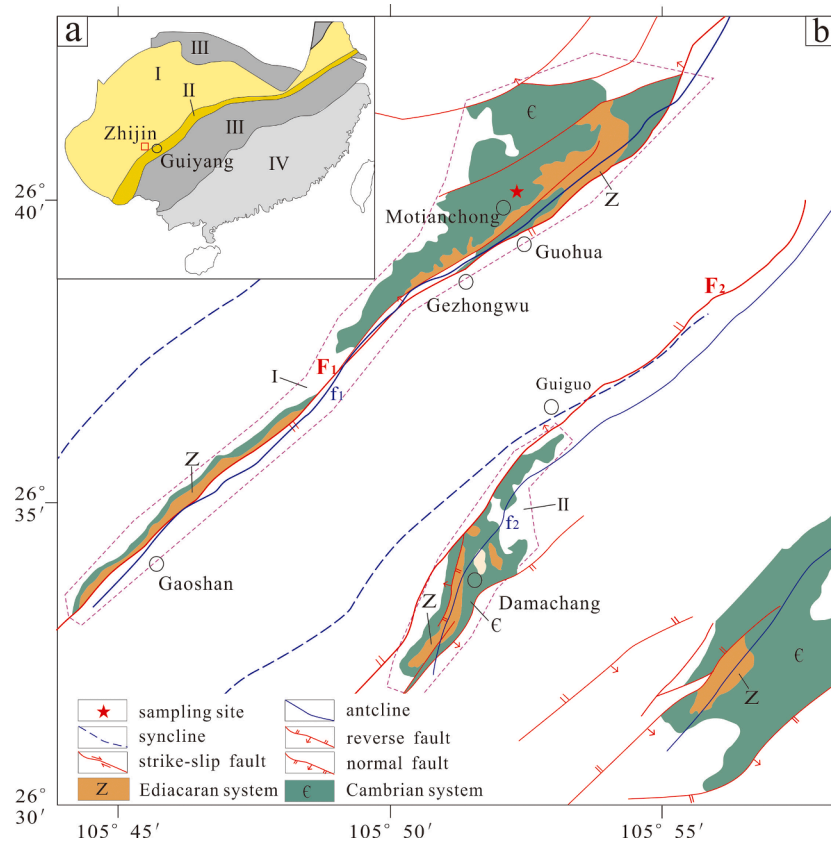


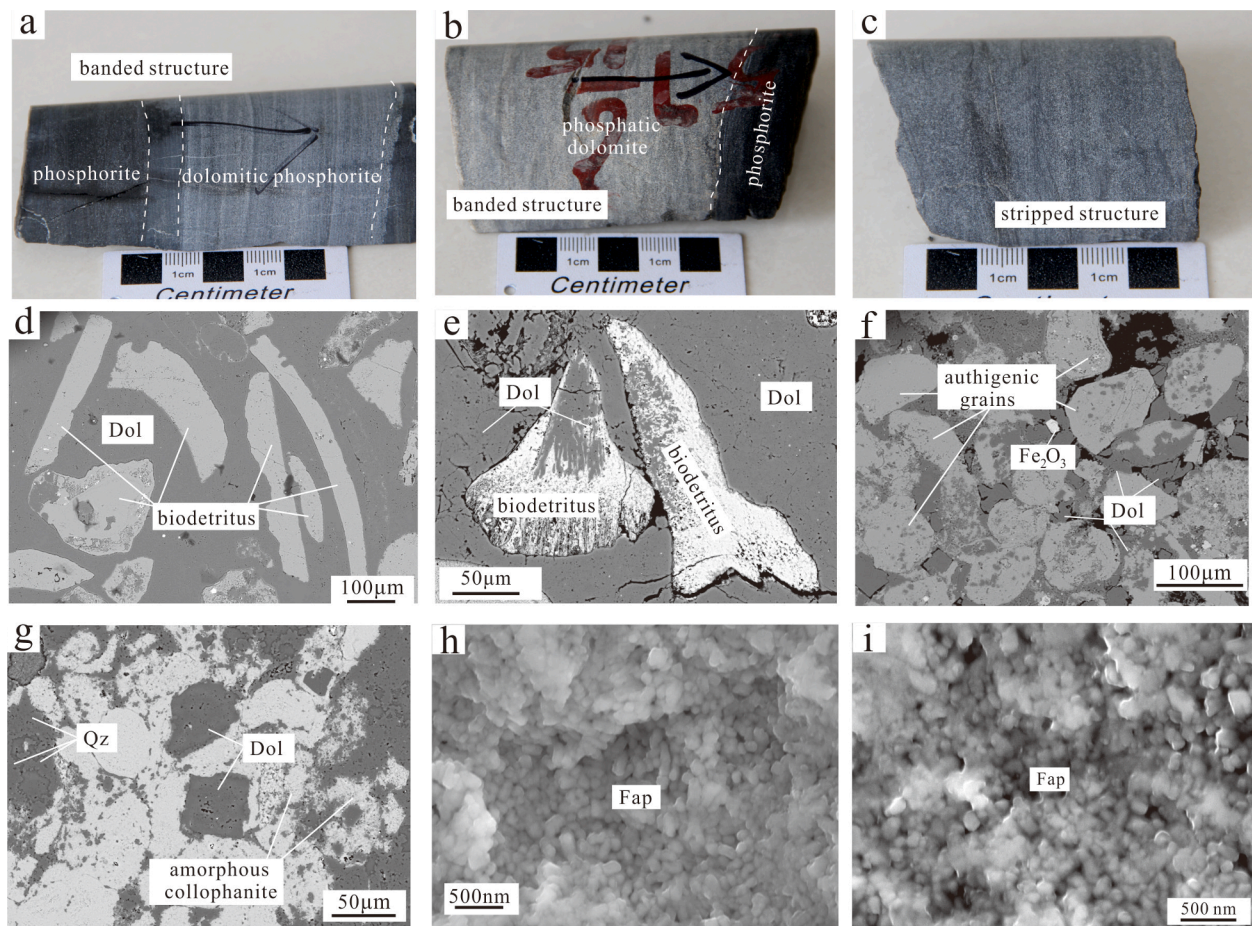
Fig. 1. (a) Simplified paleogeographic map of the Yangtze Platform in the early Cambrian, modified after Steiner et al. (2001). (b) Geological map of the Zhijin area.

including the Neoproterozoic, Cambrian, Carboniferous, Jurassic, Cretaceous, Tertiary, and modern sediments. Among them, REY-enriched phosphorites formed primarily during the Paleozoic Era (Emsbo et al., 2015). For example, REY concentrations ( $\Sigma$ REY) in Upper Devonian phosphorite reach levels as high as 18000 ppm (Kato et al., 2011; Emsbo et al., 2015), higher than ion adsorption in South China (Bao and Zhao, 2008; Wang et al., 2018). Therefore, phosphorite could serve as a new REY resource and probably meet global REY demands (Kato et al., 2011; Emsbo et al., 2015).

REY mineralization and occurrence have attracted significant research attention recently. Several viewpoints have been proposed to explain REY enrichment in sedimentary phosphorite, including: (1) An upwelling paleogeographic area favors REY enrichment. On one hand, seawater and terrigenous inputs supplied sufficient sources of P and REY (Föllmi et al., 2009; Planavsky et al., 2010; Pufahl and Groat, 2017; Yang et al., 2021a), on the other hand, shallow seawater was hydrodynamically fluctuant and facilitated phosphate precipitation and REY enrichment (Chen et al., 2013; Pufahl and Groat, 2017); (2) The paleoenvironment controlled REY enrichment. It has been approved that phosphorite formation was controlled by redox conditions through “Fe redox pumping” (Canfield et al., 2007; Nelson et al., 2010). Meanwhile, Fe- and Mn- oxides preferentially adsorbed MREEs, and REY entered apatite crystals in the pore water during diagenesis (Paul et al., 2019; Yang et al., 2021b). Furthermore, temperature, sedimentary rate, and REY composition in seawater may have played a role in REY enrichment in some degree (Al-Bassam and Magna, 2018; Yang et al., 2021b); (3) Biological activities benefitted REY enrichment. Research showed that biological absorption and organic mineralization controlled the REY cycle in the seawater and pore water, in which organic materials are enriched in MREEs (Henderson, 1984; Freslon et al., 2014). REY-enriched phosphorite always contained enriched MREEs (Emsbo et al., 2015), which implied a biological contribution (Yang et al., 2021b). Furthermore, low Zn isotopic compositions and Cd isotopes similar to

modern seawater in phosphorite implied a stronger biological contribution (Fan et al., 2018; Frei et al., 2021); (4) Fe-redox pumping control not only the phosphorite formation in non-upwelling area, but also the REY enrichment of deep sea muds in the Pacific Ocean (Nelson et al., 2010; Kashiwabara et al., 2018). These processes happened along with the Fe oxide precipitation in oxic conditions and Fe dissolution and release under redox interface; (5) Diagenesis was the key factor in REY enrichment. Most recent phosphorite exhibit seawater-like REE patterns, while deviations from modern seawater REE patterns could be ascribed to diagenetic alteration (McArthur and Walsh, 1984; Shields and Stille, 2001). Reynard et al. (1999) proposed a comparison of  $(La/Yb)_N$  vs  $(La/Sm)_N$  to identify diagenetic stages, previous reports regarding Ediacaran-early Cambrian phosphorite, Permian conodont, Jurassic fish bone, and Cenozoic phosphorite suggested that REY entered apatites during early diagenesis (Lumiste et al., 2019; Yang et al., 2021b; Yang et al., 2022). Geologists found that different mechanisms affect REY enrichment differently at the macro level, but the specific roles of different metallogenic mechanisms in REY enrichment remain unclear. Furthermore, there is no consensus on the control mechanism of REY enrichment.

Phosphate minerals, previously termed colophonite, were composed of apatite nanocrystals (Liu, 2008; Liu and Zhou, 2017; Liao et al., 2019; Zhang et al., 2019; Yang et al., 2021b). Research showed that REY occurred mainly in apatite by isomorphic substitution (Huang et al., 2021; Zhang et al., 2021), while small amounts of REY were adsorbed by phosphate or clay minerals (Zhang et al., 2007) and existed as REE independent minerals (Liu, 2008). Hence, the formation of phosphate minerals recorded key information of REY enrichment. Phosphate minerals in sedimentary phosphorite include carbon-free fluorapatite (fluorapatite) and a small amount of carbonate-fluorapatite (francolite) that comprise closely-packed, nm to  $\mu$ m-scale crystals (Liu and Zhou, 2017; Liao et al., 2019; Yang et al., 2021b). Conducting a detailed study about phosphate minerals is expected to provide further evidence for



**Fig. 2.** The characteristics of hand specimens (a)–(c) and phosphate minerals (d)–(i) from the Gezhongwu Formation, early Cambrian in Zhijin phosphorite, South China. Figures (d)–(i) are backscattered electron images, collected with the signal acquisition conditions of a 20 kV voltage and a 100 nA current. (a), (b) banded phosphorite; (c) stripped phosphorite; all francolite grains were divided into (d, e) biodetritus, (g) authigenic grains, and (h) amorphous collophanite; these francolite grains were composed of tightly packed apatite nanocrystals (f, i), with six-party and granular textures and nanoscale diameters. Among the pictures, f and i are from Yang et al. (2021b). Abbreviations: Dol = dolomite; Qz = quartz; Fap = francolite.

phosphorite formation and REY enrichment. However, limited by the small diameter of apatite nano-crystals, fine mineralogical and geochemical characteristics were difficult to obtain, severely restricting the understanding of apatite formation and REY enrichment.

The early Cambrian era was an important phosphorus-forming period (Cook, 1992), which occurred coevally with localized REY-rich phosphorite in Zhijin, South China. The Zhijin phosphorite has been studied extensively for its sedimentary, petrographic, and geochemical characteristics (Zhang et al., 2003; Liu, 2008; Chen et al., 2010), and many fine mineralogical studies were conducted in recent years (Liu et al., 2020; Yang et al., 2021b; Zhang et al., 2021; He et al., 2022; Yang et al., 2022). Previous studies provided significant information, which allows Zhijin to serve as an ideal objective for analyzing microscale apatite formation and REY enrichment.

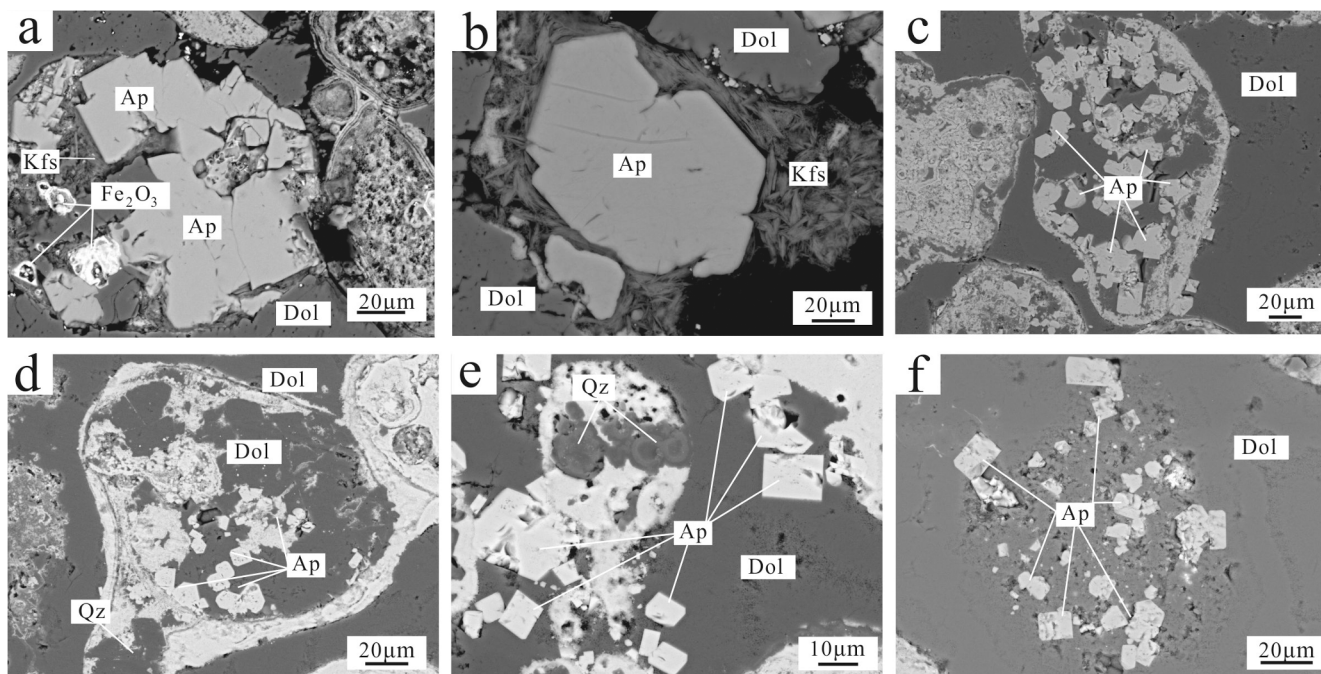
In this study, Zhijin phosphorite was examined, fine-scale mineralogical and geochemical characteristics of nano-scale apatite aggregations (collophanite) and apatite individuals were analyzed using scanning electron microscopy (SEM), electron microprobe analysis (EMPA), and laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS). The aim was to classify apatite individuals in Zhijin phosphorite, to trace the sources of apatites and REY, and to reveal the authigenic apatite formation and REY enrichment mechanism. This work provides a further understanding of REY sources and enrichment mechanism in marine sedimentary phosphorite.

## 2. Geological setting

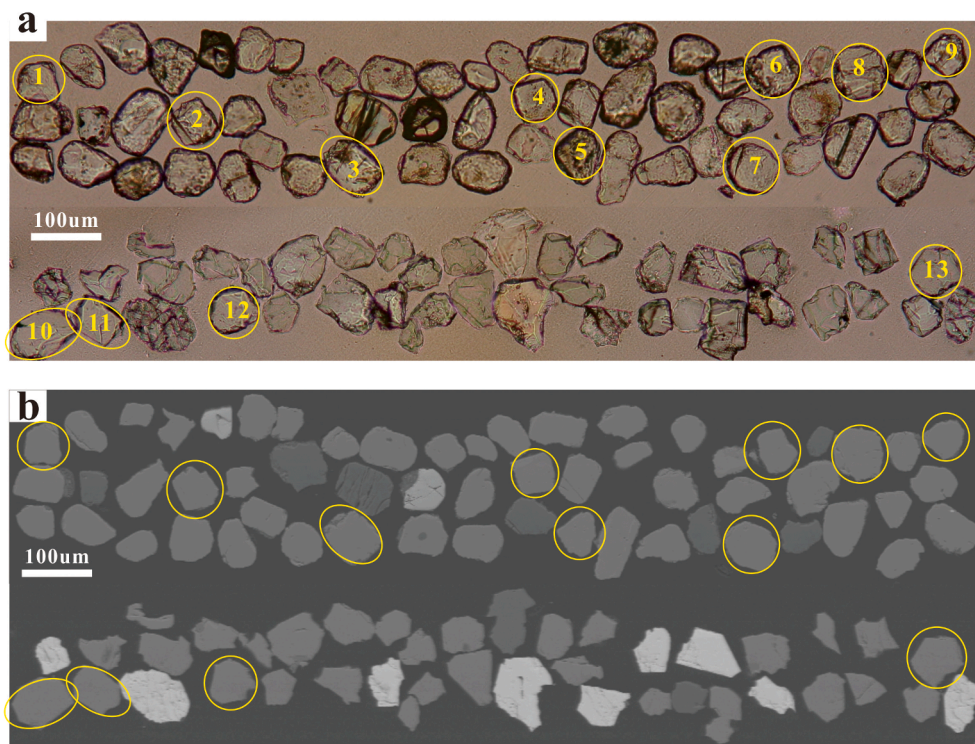
### 2.1. Paleogeography

South China underwent multiple tectonic activities after Jinningian tectonics (900–820 Ma), during which the Yangtze Block and Cathaysia Block aggregated and the Jiangnan Orogenic Belt formed (Shu, 2012; Zhang et al., 2013). Thereafter, the ancient South China Plate rapidly underwent rifting and glaciation with relatively stable tectonic movements (Shu, 2012). Along with the supercontinent fragmentation of Rodinia during the late Neoproterozoic, the Yangtze Block and Cathaysia Block began to break up (Wang and Li, 2003), which caused the paleogeographic framework of deepening water from the NW shelf towards the SE ocean basin (Liu et al., 1993; Zhang et al., 2013). In the early Cambrian, the South China Plate underwent steadier supercontinent break-ups but maintained the same paleogeographic framework of the late Neoproterozoic (Liu et al., 1993).

The early Cambrian was an important lithofacies-paleogeographic development stage in South China. During this period, the facies regions in South China were composed of shallow-water carbonate platform facies (Fig. 1a I), transitional facies (Fig. 1a II), and deep-water slope and basin facies (Fig. 1a III) from the northwest to southeast (Steiner et al., 2001). Plenty of phosphorites formed in the carbonate platforms during the early Cambrian such as Kunyang in Yunnan, Zhijin in Guizhou, and Hanyuan in Sichuan (Ye et al., 1989). Among them, Zhijin phosphorite deposit formed in the early Cambrian phase,



**Fig. 3.** Symbiotic relationship between apatite individuals with other ore minerals. Figures are collected under conditions similar with Fig. 2. (a, b) apatite individuals with larger diameters piled up with K-feldspar; (c, d) apatite individuals hosted in biodetritus; (e, f) apatite individuals distributed dispersedly and were cemented by dolomite. Abbreviations: Ap = apatite; Dol = dolomite; Qz = quartz; Kfs = K-feldspar.



**Fig. 4.** Characteristics of apatite individuals under transmitted, plane-polarized light (a) and a scanning electron microscope (b). Corresponding backscattered electron image under SEM with a 10-kV, 4.5-nA beam. All apatite individuals were selected individually. Circled numbers represent in situ analysis points, consistent with the serial numbers in Tables 2 and 3.

containing enriched REY and showing great potential REY resources. Early isotope dating suggested early Cambrian phosphorite in South China formed in  $587 \pm 17$  Ma (Cowie and Johnson, 1985),  $562 \pm 5.7$  Ma (Yang et al., 1996) by Rb-Sr and Sm-Nd isotopes, respectively. Shi (2005) subsequently conducted Sm-Nd/Rb-Sr isotopes and suggested

that Zhijin phosphorite started formation in  $541 \pm 12$  Ma. Reported re-Os isochron age of the sulfide ore layer in Niutitang Formation were confirmed to  $537 \pm 10$  Ma (Jiang et al., 2004) and  $521 \pm 5$  Ma (Xu et al., 2011). Hence, Zhijin early Cambrian phosphorite were deemed to form during  $\sim 541$ – $521$  Ma.

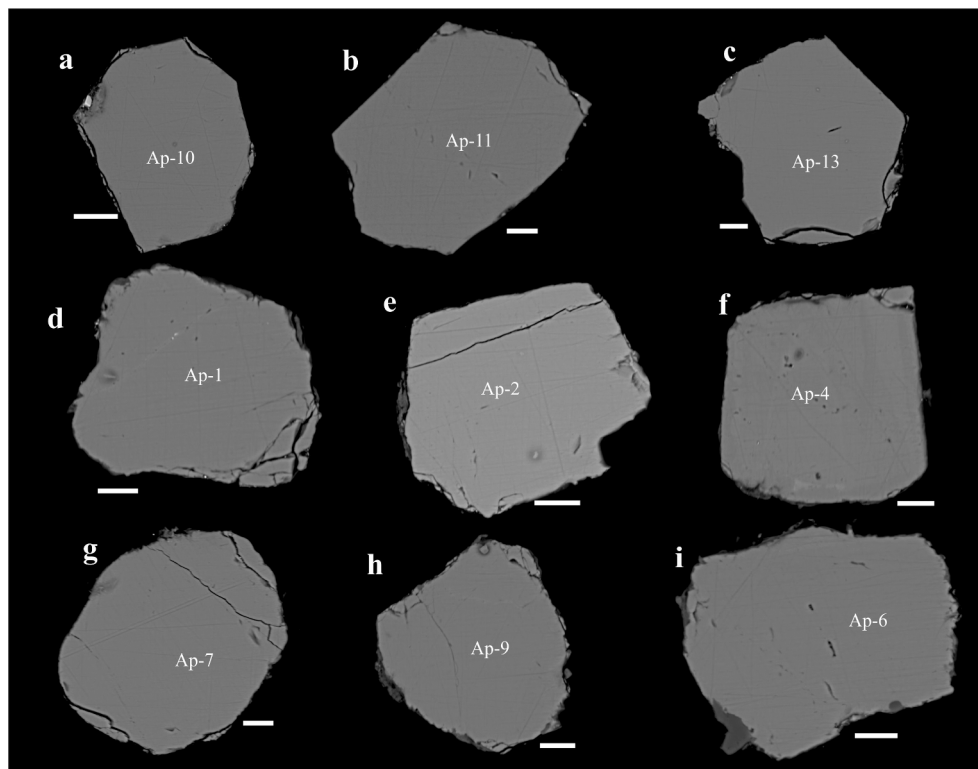


Fig. 5. Characteristics of mineralogical textures of individual apatites from the early Cambrian Zhijin phosphorite, South China. (a-c) six-party texture; (d-f) cube-like texture; (g-h) granular texture; (i) irregular texture. All images were collected via SEM, each scale bar represents 20  $\mu\text{m}$ .

Table 1

Major and trace element compositions of apatite individuals from Zhijin deposit, the Central Guizhou, South China.

| Numbers  | Li   | Na <sub>2</sub> O | Al <sub>2</sub> O <sub>3</sub> | SiO <sub>2</sub> | P <sub>2</sub> O <sub>5</sub> | K <sub>2</sub> O | CaO   | V     | Cr   | FeO  | Sr      | Ba   | Pb   | Th     | U     |
|----------|------|-------------------|--------------------------------|------------------|-------------------------------|------------------|-------|-------|------|------|---------|------|------|--------|-------|
| AP-1     | 1.92 | 0.06              | 0.05                           | 0.27             | 41.47                         | 0.01             | 57.51 | 5.43  | 0.14 | 0.27 | 401.91  | 7.19 | 3.87 | 5.73   | 4.46  |
| AP-2     | 0.09 | 0.02              | 0.04                           | 0.26             | 39.75                         | 0.01             | 59.58 | 8.19  | 0.00 | 0.03 | 412.36  | 0.58 | 4.27 | 15.15  | 29.00 |
| AP-3     | 0.21 | 0.01              | 0.00                           | 0.42             | 39.55                         | 0.00             | 59.13 | 13.39 | 0.67 | 0.03 | 496.68  | 0.95 | 5.37 | 208.21 | 92.13 |
| AP-4     | 1.36 | 0.03              | 0.05                           | 0.34             | 39.89                         | 0.03             | 58.97 | 8.45  | 0.31 | 0.10 | 316.08  | 3.89 | 3.20 | 28.61  | 7.49  |
| AP-5     | 5.18 | 0.02              | 0.19                           | 0.71             | 39.38                         | 0.07             | 58.77 | 4.25  | 0.00 | 0.20 | 328.89  | 5.11 | 3.69 | 14.32  | 3.51  |
| AP-6     | 2.59 | 0.10              | 0.19                           | 0.33             | 39.87                         | 0.06             | 58.64 | 2.42  | 1.44 | 0.25 | 147.11  | 4.84 | 3.15 | 2.47   | 12.83 |
| AP-7     | 3.24 | 0.07              | 0.00                           | 0.19             | 39.89                         | 0.00             | 58.57 | 6.46  | 0.14 | 0.60 | 360.12  | 6.90 | 4.51 | 13.55  | 7.57  |
| AP-8     | 7.59 | 0.13              | 0.17                           | 0.63             | 39.45                         | 0.06             | 58.30 | 36.83 | 0.00 | 0.10 | 287.02  | 7.09 | 3.95 | 16.63  | 3.33  |
| AP-9     | 0.37 | 0.08              | 0.00                           | 0.42             | 38.99                         | 0.00             | 58.93 | 9.94  | 0.22 | 0.05 | 529.54  | 1.08 | 5.46 | 101.30 | 33.85 |
| AP-10    | 0.00 | 0.01              | 0.00                           | 0.03             | 40.09                         | 0.00             | 59.43 | 0.07  | 0.43 | 0.01 | 2444.05 | 0.53 | 1.71 | 0.00   | 2.52  |
| AP-11    | 0.00 | 0.02              | 0.00                           | 0.01             | 40.60                         | 0.00             | 58.82 | 0.02  | 0.40 | 0.01 | 2319.59 | 1.31 | 6.22 | 0.21   | 8.48  |
| AP-12    | 0.03 | 0.02              | 0.00                           | 0.00             | 40.49                         | 0.00             | 58.98 | 0.10  | 0.00 | 0.01 | 1765.18 | 0.78 | 0.89 | 0.05   | 6.22  |
| AP-13    | 0.03 | 0.03              | 0.00                           | 0.14             | 40.24                         | 0.00             | 58.96 | 0.06  | 0.76 | 0.01 | 2106.27 | 0.90 | 9.33 | 0.03   | 20.06 |
| Mean     | 1.74 | 0.05              | 0.05                           | 0.29             | 39.97                         | 0.02             | 58.81 | 7.35  | 0.35 | 0.13 | 916.52  | 3.17 | 4.28 | 31.25  | 17.80 |
| $\delta$ | 2.38 | 0.04              | 0.08                           | 0.22             | 0.63                          | 0.03             | 0.52  | 9.85  | 0.41 | 0.17 | 879.70  | 2.74 | 2.10 | 59.63  | 24.49 |

Note: Units for Li, Na<sub>2</sub>O, Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, K<sub>2</sub>O, and CaO are wt.%, while for other elements are ppm. Mean represents arithmetic mean value,  $\delta$  represents standard deviation. The following tables also apply.

## 2.2. Ore deposit geology

### 2.2.1. Ore bodies

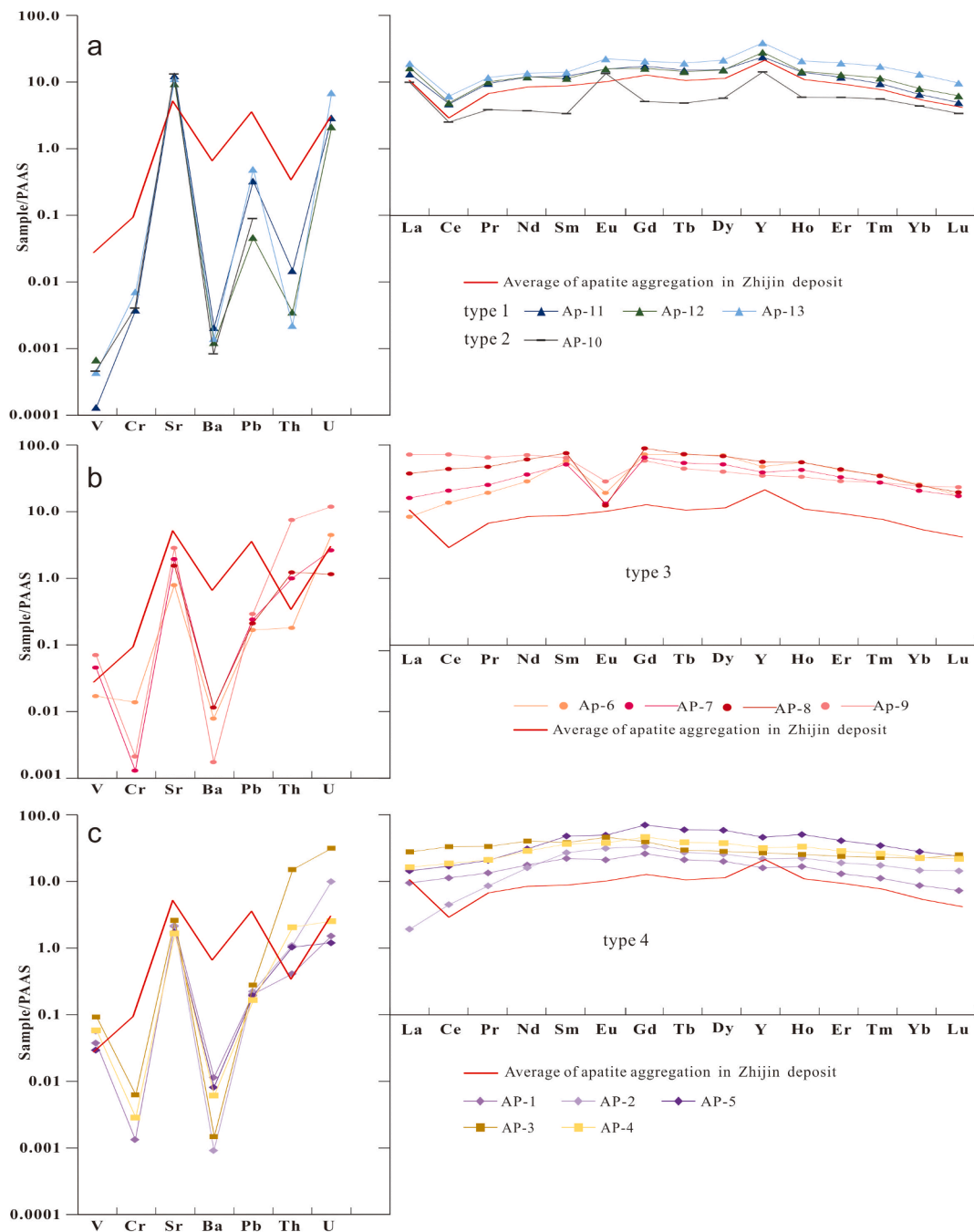
The Zhijin deposits are stratabound and spread northeast along with cores of the Guohua-Gezhongwu Fault/Anticline ( $F_1/f_1$ ) and Guiguodamachang Fault/Anticline ( $F_2/f_2$ ) (Fig. 1b). The deposits comprise an area of 333 km<sup>2</sup> and are divided into the Xinhua Mining District (I) on the northwestern side and the Damachang Mining District (II) on the southeastern side. The Xinhua Mining District includes the Daga, Motianchong, Guohua, Ganjiayakou, Gezhongwu, and Gaoshan ore blocks, which provide the most phosphate in the Zhijin deposit. The Zhijin deposit had phosphate mineral resources about 3.39 billion tons and potential REY resources (RE<sub>2</sub>O<sub>3</sub>) of nearly 3.503 million tons, with P<sub>2</sub>O<sub>5</sub> and RE<sub>2</sub>O<sub>3</sub> grades of 17.20 g/t and 0.1036 g/t, respectively (Yang

et al., 2021a).

Zhijin phosphorites are hosted in the Gezhongwu Formation, occurred in platform carbonate rocks and underlain by Dengying dolomite and overlain by Niutitang sandy shale. The Gezhongwu Formation is usually divided into the lower Gezhongwu Member and upper Gezhongwu Member, comprising dolomitic phosphorite sand striped siliceous phosphorite, respectively (Yang et al., 2021a). Biological fossils are abundant in the Zhijin deposit and include small shell fossils (SSF), macroscopic bacterial colonies, arthropods, sponges, and zooplankton (Wu et al., 1999).

### 2.2.2. Mineralogy

The texture of Zhijin phosphorites helps classify them into massive, banded, striped, earthy, and densely massive (detailed descriptions are



**Fig. 6.** PAAS-normalized rare earth element distributions of colophonite and apatite individuals with four REY types from Zhijin early Cambrian phosphorite, South China. The colophonite data came from our previous studies (Yang et al., 2021b; He et al., 2022; Yang et al., 2022), with maximum, minimum, and average value distributions. PAAS data are from Taylor and McLennan (1985).

given in Yang et al. (2021b) and Yang et al. (2022)). Among them, banded (Fig. 2a, b) and stripped (Fig. 2c) textures were widespread and typical. Minerals comprised primarily phosphates and gangue minerals including dolomite, calcite, quartz, magnetite, pyrite, K-feldspar, and muscovite. The phosphates were divided into biotritus (Fig. 2d, e), authigenic grains (Fig. 2f), and amorphous colophonite (Fig. 2g), all were aggregations of apatite nanocrystals. Biotritus have irregular textures and diameters ranging from 100 to 300  $\mu\text{m}$  and were directionally distributed, composed of hexagonal short-column fluorapatite nanocrystals (Fig. 2h). Authigenic phosphate grains have average diameters of 200–300  $\mu\text{m}$  and were characterized by elliptical, round, and spindly structures, with quartz and dolomite inclusions (Fig. 2f). Generally, authigenic grains and amorphous colophonite both are

composed of tightly packed apatite with spherulitic shapes and 20–50 nm in diameter (Fig. 2 i). We have recently reported detailed description about apatite mineral characteristics (Yang et al., 2021b; Yang et al., 2022).

### 3. Sampling and methodologies

The phosphorite samples were sampled from the Motianchong ore block in the Xinhua Mine District (Fig. 1b). The phosphorite samples were made into polished sections with thicknesses of 30  $\mu\text{m}$  for petrographic analysis, and in situ geochemical results are cited by Yang et al. (2021b) and Yang et al. (2022). Partial phosphorite samples were sieved using different diameters to select apatite individuals. Appropriate

**Table 2**  
Rare earth element compositions of apatite individuals from Zhijin deposit, the Central Guizhou, South China (ppm).

| N     | La      | Ce      | Pr     | Nd      | Sm     | Eu    | Gd     | Tb    | Dy     | Ho    | Er     | Tm    | Yb    | Lu    | Y       | ∑REY     | Ce/<br>Ce* | Eu/<br>Eu* | Y/Ho  | La/<br>Sm | La/<br>Yb |
|-------|---------|---------|--------|---------|--------|-------|--------|-------|--------|-------|--------|-------|-------|-------|---------|----------|------------|------------|-------|-----------|-----------|
| AP-1  | 339.51  | 848.76  | 111.97 | 528.81  | 115.95 | 14.00 | 114.83 | 15.25 | 82.17  | 15.72 | 35.45  | 4.20  | 22.91 | 2.95  | 404.54  | 2657.04  | 0.99       | 0.87       | 25.73 | 0.43      | 1.09      |
| AP-2  | 69.31   | 338.56  | 72.00  | 480.61  | 141.60 | 32.29 | 146.67 | 19.67 | 105.45 | 20.99 | 51.57  | 6.53  | 38.51 | 5.81  | 551.74  | 2081.32  | 0.85       | 1.04       | 26.29 | 0.07      | 0.13      |
| AP-3  | 996.59  | 2510.76 | 280.61 | 1216.05 | 202.12 | 47.62 | 174.97 | 21.54 | 117.04 | 24.01 | 65.10  | 8.70  | 58.60 | 10.16 | 682.55  | 6416.40  | 1.09       | 1.18       | 28.42 | 0.73      | 1.25      |
| AP-4  | 587.62  | 1406.59 | 176.83 | 866.06  | 193.35 | 39.27 | 205.71 | 28.07 | 156.13 | 31.34 | 78.17  | 9.85  | 59.89 | 8.89  | 808.02  | 4655.79  | 1.00       | 0.91       | 25.78 | 0.45      | 0.72      |
| AP-5  | 515.83  | 1269.61 | 172.72 | 921.76  | 250.55 | 50.99 | 308.01 | 42.74 | 240.60 | 47.21 | 110.66 | 12.90 | 73.32 | 9.57  | 1159.78 | 5186.24  | 0.96       | 0.84       | 24.57 | 0.30      | 0.52      |
| AP-6  | 294.15  | 1005.11 | 157.03 | 839.28  | 295.50 | 13.33 | 309.94 | 50.74 | 281.76 | 50.50 | 111.86 | 12.91 | 65.77 | 6.91  | 1172.11 | 4672.89  | 0.99       | 0.30       | 23.21 | 0.15      | 0.33      |
| AP-7  | 561.75  | 1522.31 | 176.83 | 1060.15 | 262.41 | 19.31 | 278.99 | 37.90 | 206.87 | 38.85 | 87.69  | 10.06 | 53.02 | 6.79  | 964.11  | 5310.62  | 1.00       | 0.23       | 24.82 | 0.32      | 0.78      |
| AP-8  | 1300.32 | 3204.36 | 383.91 | 1781.36 | 386.64 | 12.53 | 382.75 | 51.60 | 274.27 | 50.43 | 115.11 | 12.58 | 63.54 | 7.73  | 1382.40 | 9409.54  | 1.04       | 0.15       | 27.41 | 0.50      | 1.51      |
| AP-9  | 2499.60 | 5282.67 | 530.73 | 2068.16 | 329.29 | 28.67 | 250.76 | 31.23 | 160.90 | 30.63 | 76.37  | 10.07 | 62.35 | 9.23  | 861.62  | 12232.27 | 1.05       | 0.46       | 28.13 | 1.12      | 2.95      |
| AP-10 | 349.13  | 187.37  | 31.84  | 110.74  | 17.57  | 13.64 | 22.26  | 3.46  | 23.42  | 5.48  | 15.84  | 2.07  | 11.40 | 1.35  | 353.99  | 1149.57  | 0.37       | 3.15       | 64.57 | 2.93      | 2.26      |
| AP-11 | 468.86  | 352.48  | 79.51  | 356.05  | 64.48  | 16.07 | 76.04  | 10.79 | 63.11  | 13.14 | 32.03  | 3.55  | 17.08 | 1.98  | 597.39  | 2152.56  | 0.41       | 1.06       | 45.45 | 1.07      | 2.02      |
| AP-12 | 578.79  | 366.53  | 84.11  | 357.08  | 59.33  | 16.24 | 70.56  | 10.35 | 62.35  | 13.53 | 34.99  | 4.30  | 20.90 | 2.51  | 704.45  | 2385.99  | 0.37       | 1.15       | 52.06 | 1.44      | 2.04      |
| AP-13 | 673.01  | 462.35  | 97.33  | 405.43  | 73.58  | 22.92 | 90.35  | 13.83 | 87.63  | 19.37 | 52.25  | 6.43  | 34.21 | 3.89  | 977.82  | 3020.40  | 0.40       | 1.29       | 50.47 | 1.35      | 1.45      |
| Mean  | 710.34  | 1442.88 | 183.46 | 845.50  | 184.03 | 25.14 | 187.06 | 25.94 | 143.21 | 27.79 | 66.70  | 8.01  | 44.73 | 5.98  | 816.96  | 4717.74  | 0.81       | 0.97       | 34.38 | 0.83      | 1.31      |
| δ     | 621.53  | 1462.71 | 141.49 | 577.75  | 116.10 | 13.52 | 111.91 | 15.89 | 84.96  | 15.11 | 33.16  | 3.76  | 21.38 | 3.12  | 309.12  | 3173.42  | 0.30       | 0.76       | 13.71 | 0.77      | 0.83      |

Note:  $Ce/Ce^* = 2Ce_N/(La_N + Nd_N)$ ,  $Eu/Eu^* = 2Eu_N/(Sm_N + Gd_N)$  (Bau and Dulski, 1996).

apatite individuals were pasted into an orderly resin target for petrographic and in situ geochemical analyses.

### 3.1. Petrographic analyses of apatite individuals

The petrographic characteristics were analyzed via a JSM-7800F field emission scanning electron microscope (SEM) (Jeol Ltd., Japan) at the State Key Laboratory of Ore Deposit Geochemistry at the Institute of Geochemistry, Chinese Academy of Sciences. The sections and targets were sprayed with carbon, which made the section conductive, then placed in the SEM. The backscattered electron images were collected with a 10-kV, 4.5-nA beam, while EDS was conducted using a 10-kV, 4.5-nA beam.

### 3.2. In situ elemental analyses of apatite individuals

In situ trace elements of apatite, and individuals were analyzed at the State Key Laboratory of Ore Deposit Geochemistry at the Institute of Geochemistry, Chinese Academy of Sciences. The samples were ablated using a 44–60 μm laser beam and mixed with an H<sub>2</sub>/He carrier gas; that aerosol was sent into the ICP-MS for elemental signal collection. Every cyclicity of signal collection included a 30 s blank and a 50 s sample ablated signal. All elemental abundances were calibrated using NIST612 and NIST610 materials issued by the National Institute Standard Technology (NIST), using DUR as apatite and MACS as carbonate external standards. The reference levels for the external standards are available online at <https://georem.mpch-mainz.gwdg.de/>. Internal corrections were applied using elemental abundances of Ca determined using EPMA analyses to correct matrix effects between the standards and the analyzed minerals. The detection limit was at the ppb level. The signal was dealt with ICPMSDataCal (Liu et al., 2008) using multiple external standards–signal internal standard methods (Chew et al., 2016). Ce and Eu anomalies were calculated according to the following equation:  $Ce/Ce^* = 2Ce_N/(La_N + Nd_N)$ ,  $Eu/Eu^* = 2Eu_N/(Sm_N + Gd_N)$  (Bau and Dulski, 1996).

## 4. Analysis results

### 4.1. Petrographic characteristics of apatites

We have found that phosphate minerals are mainly existed in the form of fluorapatite nanocrystals constituting biotritus, authigenic grains, and amorphous colophonite (Fig. 2d-i). Most colophonites, including biotritus, containing various dolomite, quartz, and clay mineral, causing untight surface and microgranular textures (Fig. 2 d-g). Besides this, we now discovered apatite individuals with larger diameters, all apatite individuals have clean and dense transection (Fig. 3), dislike those of biotritus. Three kinds of symbiotic relationship between these apatite individuals with other ore minerals were recognized: (1) apatites with allotriomorphic texture and diameter around 80–100 μm were piled up with K-feldspar (Fig. 3a, b); (2) apatites with six-party textures and diameter about 10–30 μm occurred intensively in authigenic grains (Fig. 3c) or were hosted in biotritus (Fig. 3d); (3) apatites having cubic textures and diameter in 10–20 μm distributed dispersedly among and were cemented by dolomite (Fig. 3e, f). Both kinds (2) and (3) have idiomorphic textures and were symbiosis with and cemented by dolomite.

For apatite individuals distributed dispersedly, in this study, we selected apatite individuals with diameters much larger ranging from 40 to 100 μm for petrographic and geochemical observation. The surfaces of all apatite individuals were smooth and dense (Fig. 4). Their textures were divided into: six-party texture (Fig. 5a-c) including Ap-10 to Ap-13; cube-like texture (Fig. 5d-f) including Ap-1 to Ap-5; granular texture (Fig. 5g) including Ap-7 to Ap-9; and irregular texture (Fig. 5h, i), such as Ap- 6.

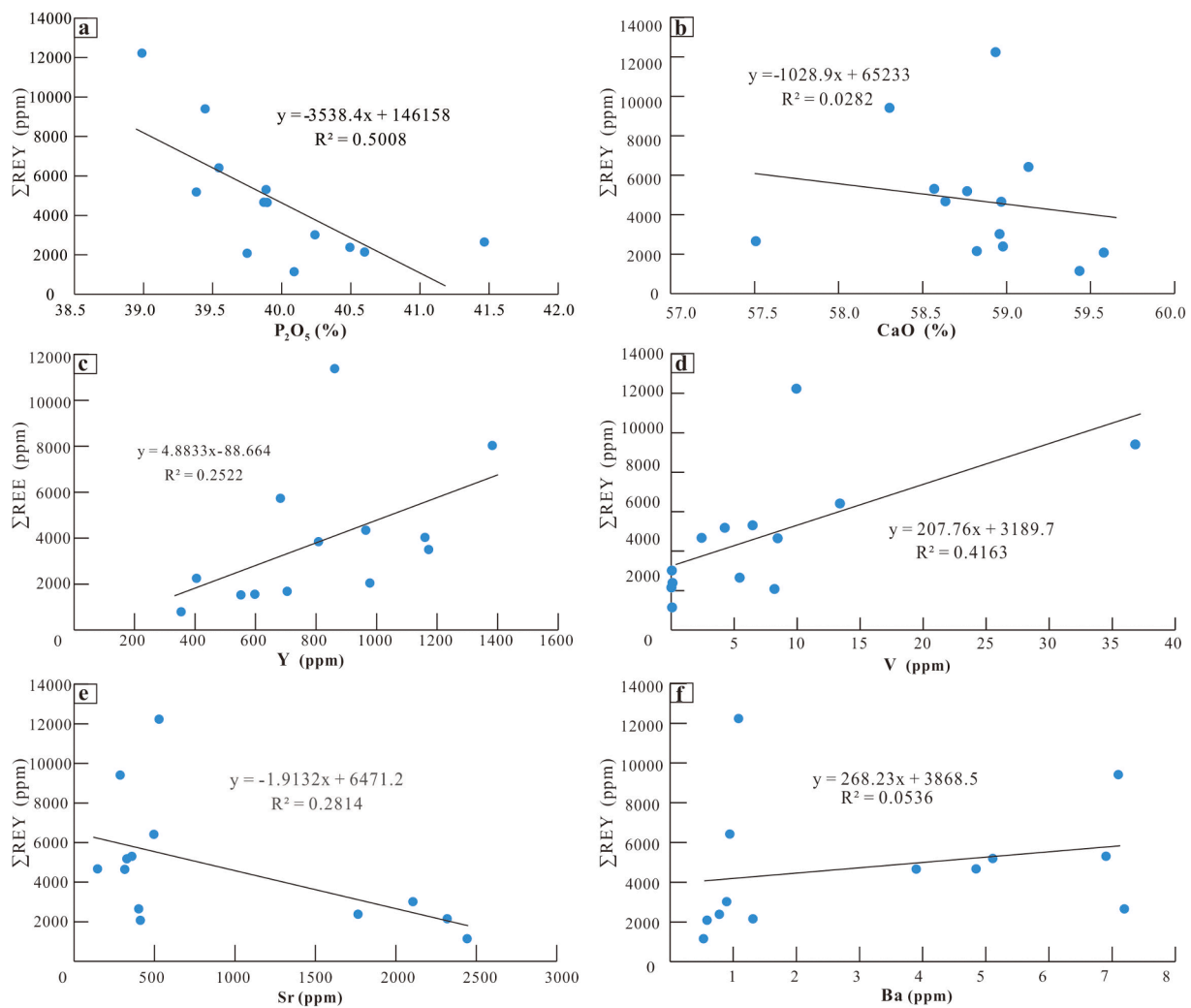


Fig. 7. Relationships between  $\Sigma$ REY and  $P_2O_5$ (a), CaO (b), Y (c), V (d), Sr (e), and Ba (f) of apatite individuals from Zhijin phosphorite.

#### 4.2. Chemical compositions of apatite individuals

In situ major and trace elements of early Cambrian apatite individuals were listed in Table 1. The apatite individuals have  $P_2O_5$  levels that range from 38.99 to 41.47 % with small variability (mean = 39.97 %, standard deviation ( $\delta$ ) = 0.63), and CaO levels of 57.51–59.58 % (mean = 58.81 and  $\delta$  = 0.52). The data show that mean values of V, Cr, FeO, Sr, Ba, Pb, Th, and U were 7.35 ppm ( $\delta$  = 9.85), 0.35 ppm ( $\delta$  = 0.41), 0.13 ppm ( $\delta$  = 0.17), 916.52 ppm ( $\delta$  = 879.70), 3.17 ppm ( $\delta$  = 2.74), 4.28 ppm ( $\delta$  = 2.10), 31.25 ppm ( $\delta$  = 59.63), and 17.80 ppm ( $\delta$  = 24.49), respectively. Apatites with six-party apatite individuals (Ap-10 to Ap-13) have higher  $P_2O_5$  (40.09–40.60 %) and Sr (1765.18–2444.05 ppm) concentrations but lower Li, Si, V, Ba, Pb, Th, and U concentrations than other apatite individuals (Fig. 6a).

In situ REY compositions of early Cambrian apatite individuals are listed in Table 2. The  $\Sigma$ REY of apatite individuals ranged from 1149.57 to 12232.27 ppm (mean = 4717.74), with great variability ( $\delta$  = 3172.42). The Post-Archean Australian Shale (PAAS) normalized REY patterns were divided into the following four groups (Fig. 6): (1) MREE-rich REY patterns, found in authigenic six-party apatite individuals (Ap-11 to Ap-13), characterized by weak enrichment of MREE compared to LREEs and HREEs and higher Y/Ho ratios (45.45–52.06), evident La and Y positive anomalies, and remarkably negative Ce anomalies (Fig. 6a); (2) There was one apatite individual (Ap-10) that showed MREE-rich patterns along with an obvious positive Eu anomaly, a high Y/Ho ratio (64.57), and low  $\Sigma$ REY (1149.57 ppm) (Fig. 6a); (3) right-inclined

to MREE enrichment patterns characterized by clear depletion of HREE and weak to remarkable LREE depletions; these apatite individuals (Ap-6, 7, 8, and 9) have notably negative Eu anomalies, low Y/Ho ratios (23.21–28.13), and high  $\Sigma$ REY (4672.89–12232.27 ppm) (Fig. 6b); (4) “bell-shaped” to flat REY patterns (found in Ap-1, 2, and 5; Ap-3 and 4, respectively) characterized by weak MREE enrichment as compared to LREEs and HREEs, with low Y/Ho ratios of 24.57–28.42 and  $\Sigma$ REY of 2081.32–6416.40 ppm (Fig. 6c). Compared with a collophanite, apatite individuals have higher and more variable  $\Sigma$ REY. Among the REY patterns, type 1 have similar REY compositions with collophanite (Fig. 6a) (Yang et al., 2021b; He et al., 2022; Yang et al., 2022). Cerium and Eu anomalies differed significantly in different textures of apatite individuals (Table 2). Authigenic six-party apatite individuals have negative Ce anomalies, with Ce/Ce\* ratios ranging from 0.37 to 0.41. In contrast, other apatite individuals have no Ce/Ce\* ratio anomalies (0.85 to 1.09). The Eu/Eu\* ratios of all apatite individuals varied from 0.15 to 3.15, with authigenic six-party apatite individuals having a mean Eu/Eu\* of 1.17 (1.06–1.29) and irregular apatite individuals having negative Eu/Eu\* of 0.15–0.46. The correlations between  $\Sigma$ REY and other elements showed that  $\Sigma$ REY negatively correlated with  $P_2O_5$  and Sr (Fig. 7a, b), positively correlated with Y and V (Fig. 7c, d), and did not correlate with Ca and Ba (Fig. 7e, f). This differed significantly from apatite aggregations we previously reported; therein,  $\Sigma$ REY showed positive correlations with Y, weakly positive correlations with Sr and Ba, and no correlations with  $P_2O_5$ , Fe,  $SiO_2$ , or  $Na_2O$  (Yang et al., 2021a).



**Table 3**  
Trace element compositions of apatite aggregations in early Cambrian phosphorite from Zhijin deposit, the Central Guizhou, South China (ppm).

| Values   | V     | Cr     | Ni     | Cu    | Zn       | As     | Rb     | Sr      | Zr    | Ba      | Pb     | Th           | U      | La      | Ce     | Pr     | Nd     |
|----------|-------|--------|--------|-------|----------|--------|--------|---------|-------|---------|--------|--------------|--------|---------|--------|--------|--------|
| Mean     | 4.32  | 10.61  | 1.63   | 7.15  | 186.86   | 12.42  | 7.46   | 1054.38 | 5.15  | 438.97  | 72.90  | 5.12         | 9.50   | 408.36  | 235.88 | 60.83  | 274.23 |
| Max      | 46.08 | 352.52 | 26.27  | 65.43 | 10117.97 | 246.88 | 106.86 | 1495.79 | 38.24 | 1273.45 | 827.67 | 30.16        | 58.11  | 735.78  | 423.63 | 115.60 | 522.85 |
| Min      | 0.21  | 0.00   | 0.00   | 0.21  | 0.23     | 1.08   | 0.01   | 522.55  | 0.15  | 19.40   | 0.57   | 0.99         | 0.60   | 1.37.84 | 109.92 | 34.61  | 160.11 |
| $\delta$ | 7.71  | 37.12  | 2.94   | 11.49 | 929.47   | 26.51  | 18.81  | 166.82  | 7.72  | 239.70  | 134.33 | 4.35         | 7.46   | 95.82   | 65.09  | 13.64  | 60.65  |
| Values   | Sm    | Eu     | Gd     | Tb    | Dy       | Ho     | Er     | Tm      | Yb    | Lu      | Y      | $\Sigma$ REY | Ce/Ce* | Eu/Eu*  | La/Sm  | La/Yb  | Sm/Yb  |
| Mean     | 49.94 | 11.32  | 60.46  | 8.23  | 50.73    | 11.03  | 27.35  | 3.09    | 15.24 | 1.83    | 579.73 | 1798.25      | 0.31   | 0.93    | 1.22   | 16.43  | 14.06  |
| Max      | 94.59 | 19.97  | 110.98 | 15.33 | 91.10    | 18.83  | 44.72  | 4.86    | 22.17 | 2.79    | 900.46 | 3067.82      | 0.39   | 1.05    | 1.66   | 24.92  | 41.01  |
| Min      | 28.61 | 6.12   | 35.04  | 4.76  | 30.53    | 6.32   | 13.41  | 1.31    | 5.16  | 0.53    | 282.18 | 969.24       | 0.27   | 0.83    | 0.47   | 12.25  | 9.96   |
| $\delta$ | 11.25 | 2.59   | 13.12  | 1.82  | 11.04    | 2.34   | 5.64   | 0.60    | 2.86  | 0.35    | 107.66 | 377.38       | 0.04   | 0.05    | 0.20   | 1.79   | 4.09   |

Note:  $\delta$  Max means maximum value, Min means Minimum value. Data are calculated according to our previous studies (Yang et al., 2021; He et al., 2022; Yang et al., 2022).

#### 4.3. Petrographic and chemical characteristics of colophonites

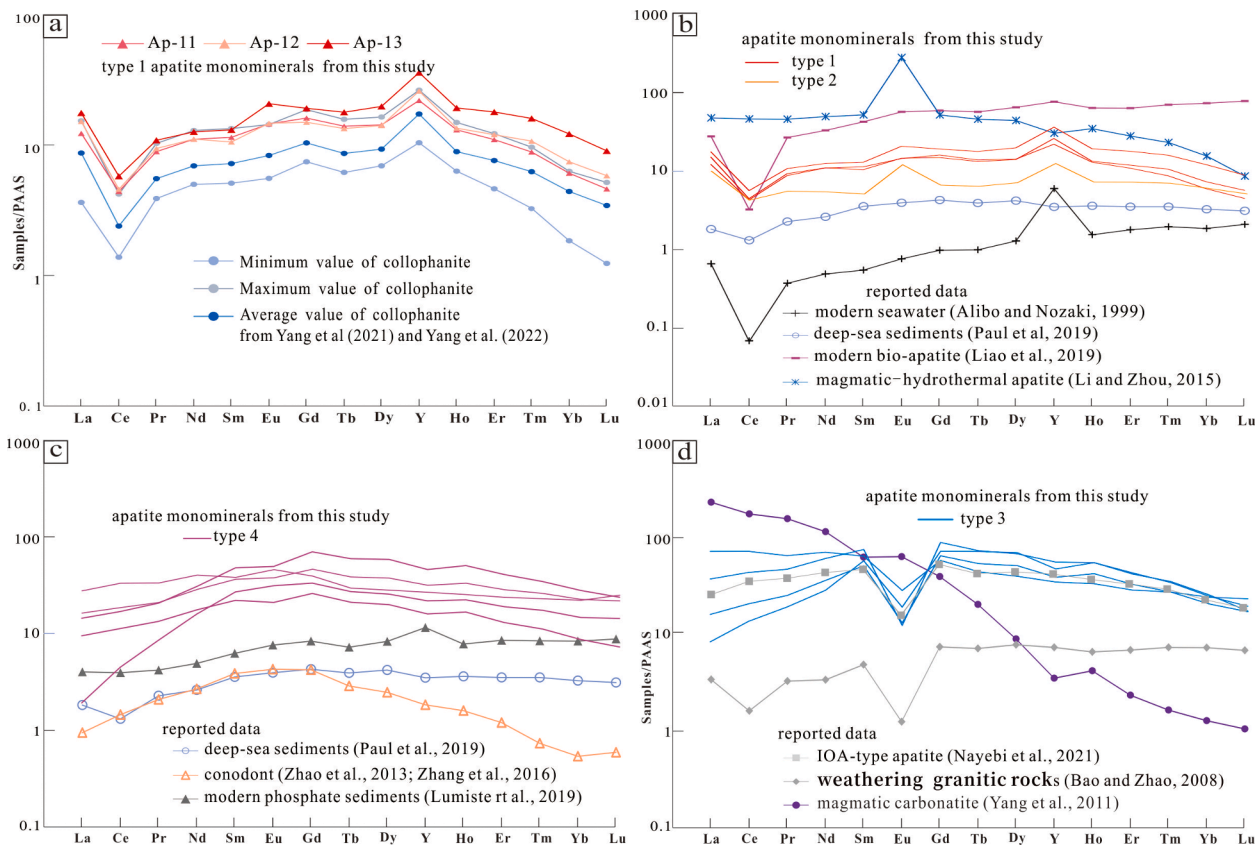
We previously reported the trace elemental compositions of colophonites, which included biotritus and authigenic grains (Yang et al., 2021b; He et al., 2022; Yang et al., 2022), and the extreme values are listed in Table 3. The trace elements of all apatite aggregations were used to compare with apatite individuals. Overall, the apatite aggregations were enriched in Sr (mean of 1054.38 ppm), Ba (mean of 438.97 ppm), Zn (mean of 186.86 ppm), and REY (mean of 1798.25 ppm). Compare with apatite individuals, apatite aggregations have higher Cr, Sr, Ba, and Pb and lower V, Th, and U and contents (mean of 10.61 ppm, 1054.38 ppm, 438.97 ppm, 72.90 ppm, 4.32 ppm, 5.12 ppm, and 9.50 ppm, respectively) (Fig. 6). All apatite aggregations have MREE-rich REY patterns as characterized by weak enrichment of MREE relative to LREEs and HREEs (Fig. 7a). In addition, apatite aggregations have positive La, Gd, and Y anomalies, remarkably negative Ce anomalies, and Y/Ho ratios ranging from 44.65 to 57.69 (Table 3), in accordance with apatites with six-party apatite individuals (Table 1).

## 5. Discussion

### 5.1. Apatite individual types and origins

Previous studies suggested that sedimentary colophonite were homogeneous and uncrystallized. However, recent research showed that colophonite comprises closely-packed, nm- $\mu$ m-scale apatites (Zhang et al., 2019; Liu and Zhou, 2020; Yang et al., 2021b), shown in Fig. 2 (g-i). Apatite individuals in Zhijin deposits have four different textures (Fig. 4). Among them, the six-party texture (Fig. 5a-c), consistent with apatite nano-crystals in colophonite, suggested they were both authigenic (discussed in Section 5.3). These cube-like textures belong to the hypidiomorphic texture (Fig. 5d-f), whereas granular and irregular belong to the allotriomorphic texture (Fig. 5g-i). This implied that they were rounded in seawater, carried over a longer distance, or underwent a certain degree of diagenesis. As we can see in Fig. 3, biotritus, authigenic grains, and amorphous colophonite all contain apatite individuals. Even though apatite individuals were observed and researched, origins of the apatite individuals can not be identified only by the morphology as the texture of apatite has been destroyed during the mineral separation process. Nevertheless, in situ geochemistry, especially REY compositions, could provide further evidence.

Usually, sedimentary phosphorites show following four REY patterns: (a) shale-like/flat patterns characterized by small or absent Ce anomalies with little to no HREE enrichment; (b) MREE-enriched patterns with significantly negative Ce anomalies; (c) "bell-shaped" patterns with no Ce anomalies and depletion of LREE and HREE; and (d) seawater-like patterns characterized by negative Ce anomalies with varying degrees of HREE enrichment (McArthur and Walsh, 1984; Shields and Stille, 2001; Emsbo et al., 2015; Lumiste et al., 2021; Yang et al., 2022). Among them, MREE enrichment or bell-shaped patterns were widely recognized as secondary signatures rather than pristine seawater (Emsbo et al., 2015; Zhu and Jiang, 2017; Yang et al., 2022). Cerium anomalies in old phosphorites could be caused by diagenesis and only in partial cases recorded the redox conditions (Shields and Stille, 2001). Previous studies showed that both phosphorite bulk rocks and colophonites in Zhijin deposits have REY patterns with MREE-enriched and negative Ce anomalies (Zhang et al., 2003; Yang et al., 2021a; Zhang et al., 2021; He et al., 2022; Yang et al., 2022). In our study, type 1 apatite individuals have REY patterns similar to cotemporaneous colophonites (Fig. 8a, b), implying similar formation process. In addition, apatite individuals with bell- to flat- REY shapes were consistent with fossils, conodont, and phosphate sediments (Fig. 8c), consistent with diagenetic origin. Therefore, the apatite individuals from Zhijin phosphorites with MREE-enriched (type 1) and bell-shape to flat (type 4) REY patterns (Fig. 6a-c) potentially formed via precipitation and post-deposition. Combining the morphology (Fig. 5) and different



**Fig. 8.** Comparisons of PAAS-normalized REY distributions between apatite individuals and collophanite (a) and different geological objects (b, c, d). Apatite individuals are from this study, collophanites are from Yang et al. (2021b) and Yang et al. (2022); conodont came from Zhao et al. (2013); modern bio-apatite came from Liao et al. (2019); deep-sea sediments came from Paul et al. (2019); modern phosphate sediments came from Lumiste et al. (2019); modern seawater came from Alibo and Nozaki (1999); IOA-type apatite came from Nayeibi et al. (2021); magmatic-hydrothermal apatites came from Li and Zhou (2015) and Xiao et al. (2021); weathering granitic rocks came from Bao and Zhao (2008), and magmatic carbonatite came from Yang et al. (2011).

**Table 4a**  
Extreme values of rare earth element compositions of reported minerals, rocks, and seawater.

|                            | values | Y       | LREE    | MREE    | HREE    | ΣREE     | ΣREY     | Y/Ho   | Ce/<br>Ce* | Eu/<br>Eu* | La/<br>Sm | La/<br>Yb | Sm/<br>Yb | Data sources                            |
|----------------------------|--------|---------|---------|---------|---------|----------|----------|--------|------------|------------|-----------|-----------|-----------|---|
| conodont                   | Mean   | 50.79   | 285.03  | 38.52   | 7.33    | 330.88   | 381.67   | 31.54  | 1.00       | 1.01       | 0.30      | 2.14      | 9.88      | Zhao et al. (2013); Zhang et al. (2016) |
|                            | Max    | 451.60  | 2855.40 | 441.62  | 59.34   | 3216.09  | 3612.19  | 47.86  | 1.79       | 1.54       | 1.19      | 10.71     | 44.00     |   |
|                            | Min    | 1.20    | 5.24    | 0.60    | 0.19    | 6.03     | 7.23     | 11.11  | 0.62       | 0.67       | 0.00      | 0.06      | 1.24      |   |
|                            | δ      | 69.92   | 395.66  | 59.36   | 9.86    | 457.50   | 523.82   | 4.68   | 0.21       | 0.16       | 0.18      | 1.56      | 8.26      |   |
| modern bio-apatite         | Mean   | 2094.71 | 2888.86 | 677.43  | 520.72  | 4087.01  | 6181.72  | 33.10  | 1.10       | 1.10       | 0.69      | 0.34      | 0.51      | Liao et al. (2019)                      |
|                            | Max    | 6025.00 | 9358.00 | 2172.00 | 1310.70 | 12840.70 | 18865.70 | 40.00  | 0.20       | 1.29       | 0.89      | 0.69      | 1.05      |   |
|                            | Min    | 79.30   | 56.61   | 17.20   | 21.33   | 95.14    | 174.44   | 30.66  | 0.00       | 1.01       | 0.55      | 0.21      | 0.25      |   |
|                            | δ      | 1407.32 | 2266.46 | 505.57  | 310.68  | 3071.23  | 4469.62  | 1.59   | 0.05       | 0.04       | 0.08      | 0.10      | 0.18      |   |
| deep-sea sediments         | Mean   | 96.49   | 304.98  | 46.92   | 26.26   | 378.16   | 474.64   | 26.15  | 0.72       | 1.00       | 0.52      | 0.57      | 1.10      | Paul et al. (2019)                      |
|                            | Max    | 208.00  | 513.60  | 99.00   | 53.65   | 648.40   | 850.65   | 28.66  | 1.07       | 1.10       | 0.64      | 0.74      | 1.23      |   |
|                            | Min    | 51.70   | 181.00  | 24.80   | 14.33   | 223.16   | 275.16   | 24.64  | 0.27       | 0.95       | 0.44      | 0.48      | 0.96      |   |
|                            | δ      | 45.06   | 78.75   | 20.72   | 11.33   | 109.45   | 153.48   | 0.92   | 0.24       | 0.17       | 0.06      | 0.07      | 0.06      |   |
| modern phosphate sediments | Mean   | 317.75  | 712.48  | 91.61   | 64.33   | 868.42   | 1186.17  | 43.94  | 0.91       | 1.07       | 0.83      | 0.44      | 0.62      | Lumiste et al. (2019)                   |
|                            | Max    | 990.00  | 2611.93 | 339.28  | 198.88  | 3147.39  | 4123.57  | 78.00  | 1.16       | 1.74       | 2.81      | 0.68      | 1.01      |   |
|                            | Min    | 10.67   | 9.78    | 2.66    | 2.02    | 14.77    | 25.44    | 33.12  | 0.56       | 0.63       | 0.54      | 0.19      | 0.10      |   |
|                            | δ      | 268.02  | 684.67  | 85.39   | 53.69   | 821.55   | 1083.57  | 9.23   | 0.15       | 0.17       | 0.45      | 0.10      | 0.24      |   |
| modern seawater            | Mean   | 164.31  | 54.03   | 12.16   | 13.94   | 80.13    | 244.44   | 104.42 | 0.17       | 1.00       | 1.15      | 0.36      | 0.32      | Alibo and Nozaki (1999)                 |
|                            | Max    | 252.20  | 103.07  | 20.94   | 22.77   | 146.78   | 398.98   | 114.29 | 0.43       | 1.11       | 1.39      | 0.41      | 0.44      |   |
|                            | Min    | 71.10   | 22.24   | 5.44    | 4.97    | 32.77    | 104.17   | 96.48  | 0.05       | 0.94       | 0.72      | 0.31      | 0.24      |   |
|                            | δ      | 63.59   | 22.68   | 4.80    | 6.34    | 33.70    | 97.01    | 3.75   | 0.12       | 0.04       | 0.20      | 0.02      | 0.06      |   |

geological objects (Fig. 8), the apatite individuals (Ap-11, 12, 13) with authigenic textures and MREE-rich REY patterns agreed with those of collophanite (Fig. 8a), which indicated similar formation (discussed in Section 5.3). Among authigenic textures, apatite individuals with positive Eu anomalies resemble magmatic-hydrothermal apatites (Fig. 8b), indicating hydrothermal origins. Differently, apatite individuals having

type 3 REY patterns (Ap-6 to Ap-9) and hypidiomorphic to allotriomorphic textures, were consistent with IOA-type apatites and weathering magmatic rocks (Frietsch and Perdahl, 1995; Bao and Zhao, 2008; Nayeibi et al., 2021), which implied terrigenous origin (Fig. 8c).

Kon et al. (2104) proposed Ce and Eu-anomalies to distinguish REY origins in deep-sea mud in the Minami-Torishima area of Japan. Here,

**Table 4b**  
Extreme values of rare earth element compositions of reported minerals, rocks, and seawater.

|                                | Y      | LREE     | MREE      | HREE    | ∑REE     | ∑REY      | Y/Ho      | Ce/<br>Ce* | Eu/<br>Eu* | La/<br>Sm | La/Yb  | Sm/<br>Yb | Data sources                                      |
|--------------------------------|--------|----------|-----------|---------|----------|-----------|-----------|------------|------------|-----------|--------|-----------|---|
| IOA-type apatite               | Mean   | 1130.52  | 5749.90   | 491.14  | 213.95   | 6454.98   | 7585.51   | 1.08       | 0.31       | 0.44      | 1.15   | 2.35      | Nayebi et al. (2021); Frietsch and Perdahl (1995) |
|                                | Max    | 11000.00 | 42230.00  | 2829.30 | 2085.70  | 43979.90  | 46059.90  | 1.24       | 0.51       | 1.09      | 5.95   | 6.83      |   |
|                                | Min    | 132.90   | 250.40    | 52.59   | 22.59    | 325.58    | 458.48    | 25.57      | 0.94       | 0.07      | 0.08   | 0.58      |   |
| magmatic-hydrothermal apatites | Mean   | 1317.21  | 5640.59   | 417.18  | 257.55   | 6033.83   | 6784.18   | 1.70       | 0.09       | 0.30      | 0.99   | 0.92      | Li and Zhou (2015); Xiao et al. (2021)            |
|                                | Max    | 1321.15  | 9548.65   | 850.54  | 256.98   | 10656.17  | 11977.32  | 25.16      | 2.80       | 0.63      | 2.21   | 2.99      |   |
|                                | Min    | 6500.00  | 29790.00  | 3325.00 | 1127.70  | 34242.70  | 40742.70  | 37.50      | 6.13       | 1.64      | 6.87   | 5.52      |   |
| weathering granitic rocks      | Mean   | 9.44     | 60.80     | 44.42   | 0.91     | 164.11    | 209.09    | 13.67      | 1.18       | 0.07      | 0.09   | 0.34      | Bao and Zhao (2008)                               |
|                                | Max    | 4299.06  | 9677.97   | 663.08  | 231.92   | 10248.46  | 10950.75  | 4.96       | 1.35       | 0.55      | 2.30   | 1.12      |   |
|                                | Min    | 201.94   | 433.06    | 77.67   | 53.83    | 564.56    | 766.50    | 30.70      | 0.65       | 0.66      | 1.01   | 1.12      |   |
| magmatic carbonatite           | Mean   | 635.00   | 965.87    | 225.88  | 192.14   | 1131.38   | 1324.18   | 48.32      | 1.81       | 1.29      | 3.60   | 2.79      | Yang et al. (2011)                                |
|                                | Max    | 27.95    | 48.25     | 10.54   | 8.60     | 114.57    | 142.52    | 24.69      | 0.20       | 0.10      | 0.02   | 0.20      |   |
|                                | Min    | 187.91   | 343.93    | 58.20   | 52.45    | 338.48    | 381.91    | 5.94       | 0.41       | 0.43      | 1.02   | 0.85      |   |
| magmatic carbonatite           | Mean   | 96.99    | 28640.62  | 316.49  | 16.07    | 28973.18  | 29070.17  | 24.39      | 1.29       | 2.49      | 134.11 | 36.75     | Yang et al. (2011)                                |
|                                | Max    | 506.00   | 192206.00 | 2612.00 | 70.96    | 194888.96 | 195394.96 | 47.50      | 1.79       | 8.98      | 552.96 | 96.88     |   |
|                                | Min    | 2.85     | 10.01     | 0.74    | 0.43     | 11.18     | 14.03     | 17.26      | 0.99       | 0.23      | 0.17   | 0.62      |   |
| δ                              | 126.32 | 51897.89 | 600.00    | 17.04   | 52478.38 | 52589.57  | 8.35      | 0.07       | 0.24       | 2.43      | 170.14 | 36.08     |   |

Note: Initial rare earth elements in the references are normalized by PAAS (Taylor and McLennan, 1985) in this table. LREE = La + Ce + Pr + Nd + Sm; MREE = Eu + Gd + Tb + Dy; HREE = Ho + Er + Tm + Yb + Lu; Ce/Ce\* = 2Ce<sub>N</sub>/(La<sub>N</sub> + Nd<sub>N</sub>), Eu/Eu\* = 2Eu<sub>N</sub>/(Sm<sub>N</sub> + Gd<sub>N</sub>) (Bau and Dulski, 1996).

we recalculated Ce and Eu-anomalies of Zhijin collophanites and other reported geologic objects using PAAS normalization (Table 4). In the Ce vs Eu diagram (Fig. 9a), sediments, including conodont, modern phosphate sediments, and deep-sea mud, generally had Eu/Eu\* ratios ~ 1.0 and variable Ce/Ce\* ratios (~0.1–1.79) (Zhao et al., 2013; Zhang et al., 2016; Lumiste et al., 2019; Paul et al., 2019). Furthermore, modern seawater, as well as modern bio-apatite, has an Eu/Eu\* ratio ~ 1.0 and lower Ce/Ce\* (0.1–0.4) ratios (Alibo and Nozaki, 1999; Liao et al., 2019). However, iron-oxide-apatite (IOA)-type apatites and weathering magmatic rocks have lower Eu/Eu\* ratios ~ 0.3 and variable Ce/Ce\* ratios (~0.65–1.81) (Bao and Zhao, 2008; Nayebi et al., 2021). In addition, hydrothermal apatite has Eu/Eu\* ratios ~ 1.0 and extremely high Ce/Ce\* ratio of 6.13 (Li and Zhou, 2018; Xiao et al., 2021). In Ce/Ce\* vs Eu/Eu\* discrimination figures, Zhijin collophanites plotted in the seawater area (Fig. 9b), indicated the seawater autogenesis. The apatite individuals were plotted primarily in three areas: IOA-type ore, fossil apatite, and seawater (Fig. 9b). These were consistent with REY patterns (Fig. 8): apatites (Ap-11, 12, 13) with REY patterns similar to collophanite located close to collophanite; apatites having flat and bell-shaped REY patterns located at conodont and fossil area; apatites with REY patterns similar to IOA-type also located in the IOA ore area. In addition, Ap-10 had high positive Eu anomalies due to hydrothermal influences.

The REY patterns and Ce/Ce\* vs Eu/Eu\* figures suggested that the apatite individuals were divided into: authigenic apatites (AP-11, 12, and 13), diagenetic apatites (Ap-1 to Ap-5), terrigenous apatites (Ap-6 to Ap-9), and hydrothermally altered apatites (Ap-10). It implied that authigenic/diagenetic apatites might undergo authigenic formation and diagenetic alteration during deposition while other apatite individuals come from weathering of terrestrial volcanic rocks. Even though it is not quite distinguishable whether the biotritus were mistaken for apatite individuals, slightly different REY compositions among these apatite individuals could be discussed to support the REY enrichment mechanism. Hence, apatite individuals could trace the sources of apatite and the REY, while authigenic/diagenetic apatites maintained formation and REY enrichment information of marine sedimentary apatites.

### 5.2. REY sources

Yttrium has a similar outer electron arrangement and radius with REEs (especially Ho), displaying highly coherent behavior. However, Y fractionates with REEs under certain conditions. The variation of Y with respect to Ho may provide additional geochemical information, such as to distinguish the sources (Bau and Dulski, 1995; Bau et al., 1995). In magmatic systems, neither partial melting nor fractional crystallization significantly fractionated Y from Ho, hence the apatites from magmatic systems had Y/Ho ratios close to chondrite (28) (Bau, 1996). Modern seawater have higher Y/Ho ratios ranging from 96 to 114 (Alibo and Nozaki, 1999). In addition, hydrothermal Y/Ho ratios have two groups of data, even though hydrothermal vein fluorites had variable Y/Ho ratios (up to 200), fluorites derived from and deposited near igneous rocks have Y/Ho ratios close to igneous source-rocks (Bau and Dulski, 1995), which was true in magmatic-hydrothermal apatites (~25) (Fig. 10) (Li and Zhou, 2015; Xiao et al., 2021). In the Y/Ho vs ∑REY figure (Fig. 10), authigenic apatites were located around collophanite and modern phosphate sediment areas (Yang et al., 2021b; He et al., 2022; Yang et al., 2022), while diagenetic apatites were close to bio-apatite and conodont (Zhao et al., 2013; Zhang et al., 2016; Liao et al., 2019) with lower Y/Ho and higher ∑REY. Terrestrial apatite individuals were located close to IOA-type apatites, magmatic and hydrothermal, and magmatic carbonite areas (Frietsch and Perdahl, 1995; Yang et al., 2011; Nayebi et al., 2021), while hydrothermal altered apatites located independently with lower ∑REY (Ap-10). These characteristics are consistent with REY PAAS-normalized patterns (Fig. 8) and Ce-Eu binary figure (Fig. 9), supporting the apatite classification. REY compositions and Y/Ho ratios of terrigenous clastic apatite individuals

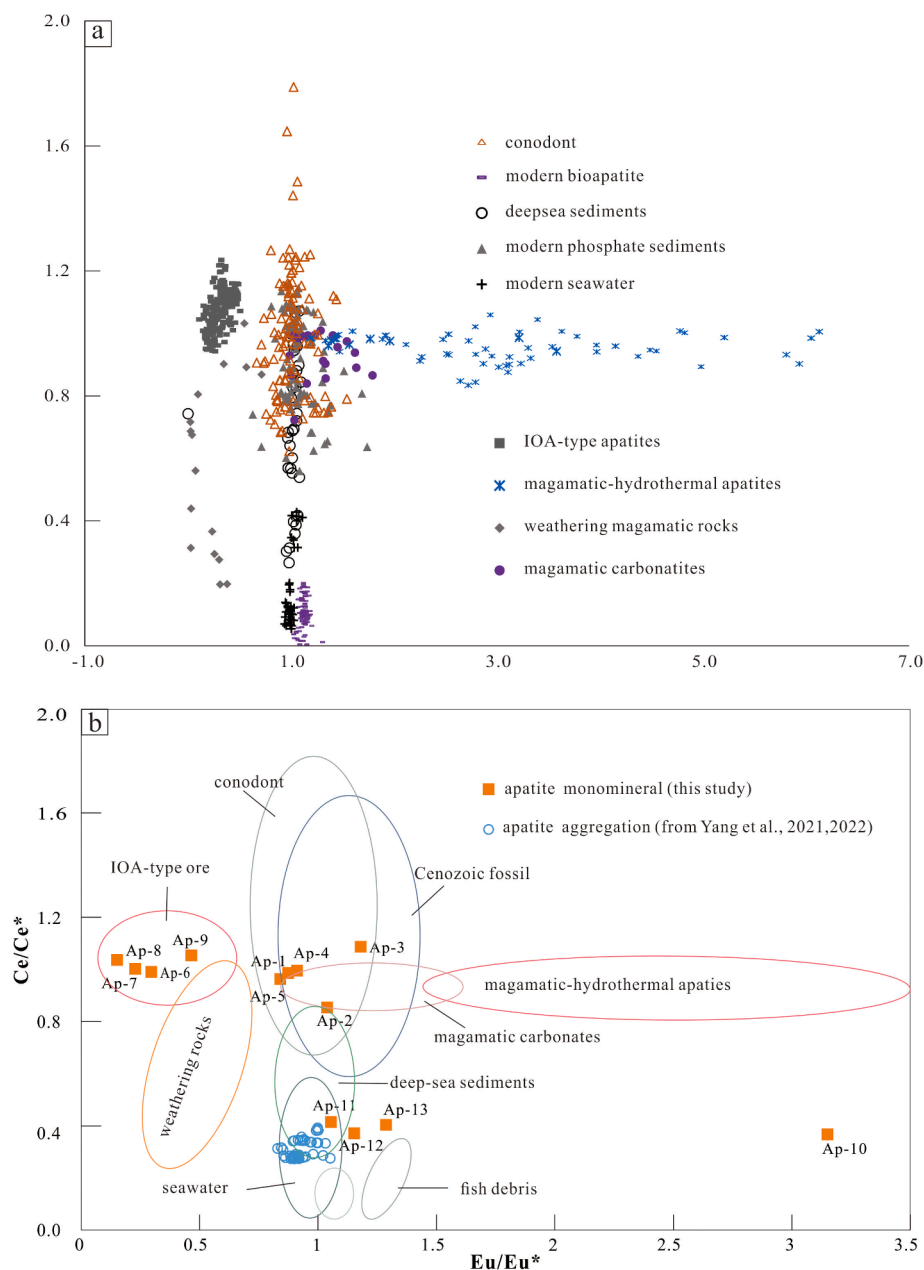


Fig. 9. Ce/Ce\* and Eu/Eu\* binary figure of different geologic objects (a) and Ce/Ce\*-Eu/Eu\* discriminant figure of the apatite sources (b). Base map and Ce/Ce\* and Eu/Eu\* according to Kon et al. (2104) and references therein, normalized by PAAS, data of different geological objects came same the same as data in Fig. 8.

indicated they came from terrestrial volcanic rock weathering and retained the initial Y/Ho ratios. However, authigenic/diagenetic apatites recoded the geochemistry of seawater during formation and diagenetic information.

The weathering of terrestrial volcanic rocks not only carried clastic apatite, but also provided high REY in seawater, as evidenced from Y/Ho vs REY of bulk phosphorite, collophanite, and apatite individuals. Previous studies showed that seawater had higher Y/Ho ratios, whereas terrestrial igneous rocks and weathering granites have lower Y/Ho (Alibo and Nozaki, 1999; Bao and Zhao, 2008; Yang et al., 2011). Hence the negative relationships between Y/Ho and  $\sum$ REY in phosphorite (Yang et al., 2021a) indicated that terrestrial inputs increased the  $\sum$ REY in seawater. This was consistent with apatite individuals, which usually have lower Y/Ho ratios and higher  $\sum$ REY (Fig. 9), especially for terrigenous apatites with high  $\sum$ REY levels between 4655 and 12232 ppm. Bau et al. (1995) reported that Y/Ho ratios ranged from 87 to 119 in South Pacific seawater at varying depths, consistent with data from

the Boso Peninsula, Japan (96–114) (Alibo and Nozaki, 1999). The authigenic phosphorite rocks and collophanite have Y/Ho ratios between igneous rocks and seawater, which indicated an input mixture of terrestrial rocks and seawater and were consistent with previous studies (Yang et al., 2021a; He et al., 2022).

### 5.3. Formation of authigenic apatites

Previous studies reported REY in collophanites to reveal the apatite formation and REY enrichments. However, collophanites contain impurities including carbonates, quartz, organic matter, and other minerals (Fig. 2d-g) that inhibited the geochemical characteristics of phosphate minerals. Furthermore, the small diameters of apatite nanocrystals in collophanites limited a deeper understanding of the fine process of apatite formation and REY enrichment. Here, we analyzed the geochemistry of authigenic apatite individuals, compared to collophanites, to discuss the formation of authigenic apatites and REY

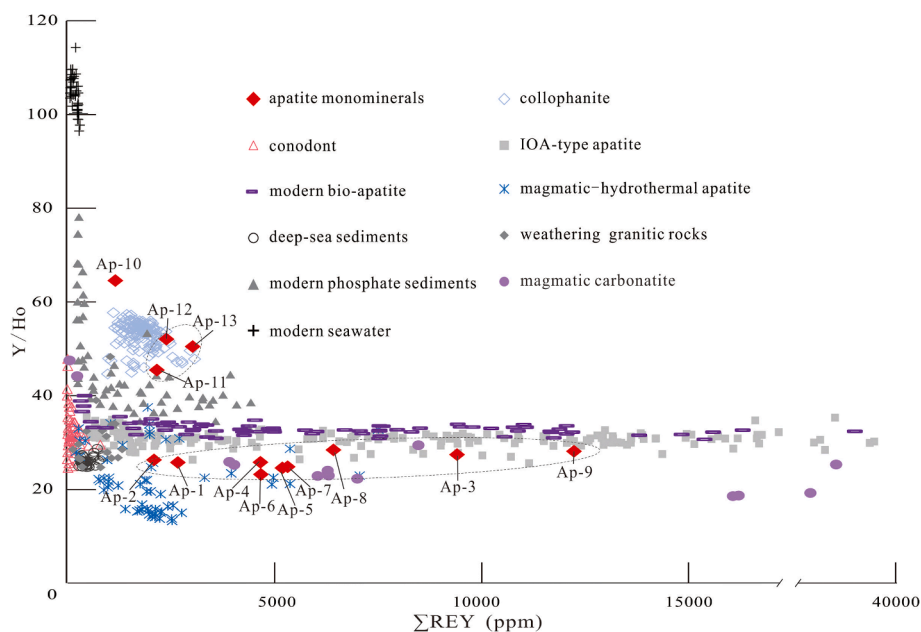


Fig. 10. Relationships between  $\Sigma$ REY and Y/Ho for different rocks, sediments, and seawater. Apatite individuals are from this study, collophanites came from Yang et al. (2021b) and Yang et al. (2022), and all reported data sources are the same as Fig. 8.

enrichment mechanisms. Apatite individuals were divided in four ways (Section 5.1), among them, authigenic/diagenetic apatites are used to discuss the formation mechanism and REY enrichment.

Previous studies showed that phosphogenesis went through prolonged diagenesis after precipitation (Shields and Stille, 2001; Chen et al., 2003), during which the phosphate mineral formed and the mineral phase transformed (Oxmann and Schwendenmann, 2014; Liao et al., 2019). Along with burial and diagenesis,  $\text{CO}_2$  in apatites disappeared and high-carbon apatites transformed into low-carbon or carbon-free apatites (Liu, 1989). X-ray diffraction (XRD) showed that phosphate minerals in Zhijin deposits were carbon-free fluorapatite with small amounts of carbonate-fluorapatite, which indicated a diagenetic transformation (Liu, 2008; Yang et al., 2021b). Even though the content of selected individuals was not enough to analyze their XRD characteristics, authigenic apatite individuals have larger diameters (Figs. 3, 4, 5) than apatite nanocrystals in collophanites (Fig. 2h, i), resulting from that phosphate minerals transformed and grew up during diagenesis.

The geochemical characteristics of collophanites and apatite individuals were compared to discuss apatite formation, especially diagenesis (Shields and Stille, 2001; Lumiste et al., 2019; Lumiste et al., 2021; Yang et al., 2022). Lécuyer et al. (2004) proposed that a  $(\text{La}/\text{Sm})_N > 0.3$  represents initial conditions or slight alterations, while  $(\text{La}/\text{Sm})_N < 0.3$  come from the addition of REEs substituted into the crystals. Reynard et al. (1999) proposed a combined  $(\text{La}/\text{Yb})_N$  and  $(\text{La}/\text{Sm})_N$  plot to distinguish the diagenetic stage and mechanisms of REE incorporation in apatite. Previous reports showed that Zhijin collophanites have higher  $(\text{La}/\text{Yb})_N$  (2.0–3.2) and  $(\text{La}/\text{Sm})_N$  ( $>0.5$ ), whereas authigenic apatite individuals have  $(\text{La}/\text{Sm})_N$  (1.1–1.4) and  $(\text{La}/\text{Yb})_N$  (1.4–2.1) similar to collophanite, which indicated early diagenetic adsorption instead of late substitution (Yang et al., 2021b; Zhang et al., 2021). However, diagenetic apatites have lower  $(\text{La}/\text{Sm})_N$  and  $(\text{La}/\text{Yb})_N$  ratios (0.07–0.73 and 0.13–1.25, respectively), which indicated they underwent stronger diagenesis than authigenic apatite. Other study showed that diagenesis caused REY redistribution and increased  $\Sigma$ REY and Ce levels (Shields and Stille, 2001). Authigenic apatite individuals had lower  $\Sigma$ REY, significantly negative Ce, and positive Y abnormalities relative to diagenetic apatites (Fig. 6a, c), this supported that diagenetic apatites enduring stronger diagenesis had higher REY and modified REY distributions, consistent with the Ce-Eu discriminant figure (Fig. 9b).

Meanwhile, Eu fractionated from other REY occurs via: (1) hydrothermal activities, (2) redox conditions, and (3) diagenetic modification. High temperatures ( $>250$  °C), anoxic conditions, and diagenesis can increase Eu in apatites (Olivarez and Owen, 1991; Shields and Stille, 2001; Kidder et al., 2003; Deng et al., 2017). In Zhijin phosphorite, the Ce/Ce\* ratios of collophanites (0.27–0.39) and apatite individuals (0.37–0.41) both indicated oxic conditions (Fig. 9), consistent with other isotopic indexes, such as Fe and Mo (Wen et al., 2010; Fan H.F. et al., 2016; Yang et al., 2021a). In addition, only noteworthy positive Eu anomalies were influenced by hydrothermal fluids, such as Ap-10. Hence, apatite individuals with weakly Eu/Eu\* ratios indicated they underwent more remarkable diagenesis than collophanites during growth, rather than resulted from redox conditions (Fig. 9 b).

#### 5.4. REY enrichment mechanisms in authigenic apatites

Ignoring the REY sources and secular variations of  $\Sigma$ REY in ocean chemistry, REY enrichment in Zhijin phosphorite could be controlled by the paleogeographic environment, Fe-redox pumping, biological absorption, and prolonged diagenesis (Kashiwabara et al., 2018; Yang et al., 2021b; Zhang et al., 2022). In this study, authigenic/diagenetic apatite individuals have much higher REY than collophanites (Fig. 8a), implying that there must be special factors caused apatite individuals to be enriched with REY. Firstly, formed in the same paleogeographic location, distinction between collophanites and authigenic apatite individuals could not be interpreted as paleogeographic environment. Secondly, in the definite case that early Cambrian seawater in South China was oxic (Wen et al., 2010; Fan H.F. et al., 2016; Yang et al., 2021a), organic matter or Fe/Mn oxides/hydroxides served as efficient traps for REY. Previous studies proposed that MREEs were preferentially adsorbed by organic matter or Fe/Mn oxides/hydroxides, then were released into pore water (Wright et al., 1987; Kidder and Eddyilek, 1994; Felitsyn and Morad, 2002). Thereafter, during diagenesis, REY was redistributed in pore water and the phosphate mineral phase, producing MREE-enrichment to bell-shaped patterns (Emsbo et al., 2015; Al-Bassam and Magna, 2018; Lumiste et al., 2021). These characteristics were found both in Zhijin collophanites and apatite individuals (Fig. 6), which indicated that preferential adsorption of MREEs and diagenetic modification controlled the REY enrichment. However, compared to

collophanites, apatite individuals in the Zhijin deposit were larger and had higher  $\sum\text{REY}$ ; in addition, diagenetic apatite individuals had higher  $\sum\text{REY}$  (Tables 1, 3, Fig. 6). As discussed in Section 5.3, higher  $\text{Eu}/\text{Eu}^*$  suggested more remarkable diagenesis of apatite individuals than apatite nanocrystals in collophanites. Hence, we conclude that during diagenesis, REY entered into apatite and enriched along with their growth.

## 6. Conclusions

(1) In early Cambrian Zhijin phosphorite deposits, REY-rich phosphate minerals developed mainly as collophanites, an aggregation of packed apatite nano-crystals and a small amount of apatite individuals. The apatite individuals have diameters between 40 and 100  $\mu\text{m}$  and were divided into authigenic/diagenetic apatites, terrigenous clastic apatites, and hydrothermally altered apatites.

(2) In the South China ocean during the early Cambrian, weathering of terrestrial IOA deposits and magmatic-hydrothermal rocks carried clastic apatite and provided high REY into seawater. Furthermore, REY also came from deep seawater, resulting in mixed Y/Ho ratios in collophanites and authigenic/diagenetic apatite individuals.

(3) Comparing the geochemical characteristics of collophanites, authigenic apatite individuals showed similar REY patterns, implying a similar formation process during which diagenetic modification played a vital role. Authigenic/diagenetic apatite individuals have larger diameters, higher  $\text{Ce}/\text{Ce}^*$  and  $\text{Eu}/\text{Eu}^*$  ratios, and lower  $(\text{La}/\text{Yb})_N$  and  $(\text{La}/\text{Sm})_N$  relative to collophanites, which suggested that apatite individuals underwent stronger diagenesis during formation.

(4) Preferential adsorption of MREEs and diagenetic modification might control REY enrichment in authigenic/diagenetic apatites, stronger diagenesis led to higher REY in larger apatite individuals. Organic matter or Fe/Mn oxides/hydroxides preferentially adsorbed REEs and subsequently released into pore water. During diagenesis, phosphate minerals transformed and grew, along with REY entering into apatite crystals.

## Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

## Data availability

Data will be made available on request.

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