

Full Length Article

Fate of mercury in two CFB utility boilers with different fueled coals and air pollution control devices



Xinyu Li^{a,b}, Zhonggen Li^{a,*}, Chengcheng Fu^c, Li Tang^a, Ji Chen^a, Tingting Wu^a, Che-Jen Lin^{a,d,e}, Xinbin Feng^{a,*}, Xuewu Fu^a

^a State Key Laboratory of Environmental Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences, Guiyang 550081, China

^b University of Chinese Academy of Sciences, Beijing 100049, China

^c Guizhou Environmental Monitoring Center, Guiyang 550081, China

^d Center for Advances in Water and Air Quality, Lamar University, Beaumont, TX 77710, USA

^e Department of Civil and Environmental Engineering, Lamar University, Beaumont, TX 77710, USA

ARTICLE INFO

Keywords:

Circulating fluidized bed (CFB) boiler

Mercury

Speciation

Removal efficiency

Emission factor

ABSTRACT

Mercury (Hg) emitted from two circulating fluidized bed (CFB) utility boilers (150 MW, P#1 and 300 MW, P#2) in Guizhou province, Southwest China, was characterized. P#1 used anthracite coal and was equipped with in-furnace desulfurization (IFD) and electrostatic precipitator (ESP). P#2 co-burned gangue and coal slime and was equipped with selective non-catalytic reduction (SNCR), ESP, fabric filter (FF) and limestone gypsum wet flue gas desulfurization (WFGD). Flue gas samples from the inlets and outlets of air pollution control devices (APCDs) in the two plants were analyzed using the Ontario Hydro Method (OHM). Solid samples including feed coal, limestone, bottom ash, fly ash and gypsum were also collected and analyzed. Hg removal efficiency by the APCDs and Hg emission factors (MEFs) were estimated. Total Hg (THg) in the flue gas at the inlet of ESP are $32.3 \mu\text{g}\cdot\text{m}^{-3}$ and $15.3 \mu\text{g}\cdot\text{m}^{-3}$ for P#1 and P#2, respectively, both dominated (95.11–99.26%) by the particulate form (Hg^{p}). THg decreased significantly to $0.19\text{--}0.24 \mu\text{g}\cdot\text{m}^{-3}$ at the final APCDs outlets of these two power plants, yielding an overall Hg removal efficiency of 98.53–99.41% for THg. Mass balance analysis indicated that > 97% of Hg in coals ended up in captured fly ash. MEFs of the two tested power plants are relatively lower than the values typically found in CFB and pulverized coal (PC) utility boiler power plants. The emission data of Hg from CFB coal-fired power plants should be updated and attention should be directed to the secondary emission/release of Hg from the captured fly ash.

1. Introduction

Hg, with its toxicity, environmental persistence, atmospheric transport and bioaccumulation, has been listed as a priority pollutant worldwide [1,2]. Anthropogenic Hg emissions exceeded that of natural sources gradually and became the main origin of Hg in the global environment after the industrial revolution [3–5]. The massive Hg emission into the atmosphere has caused adverse effects on the ecological environment [1], such as the obvious accumulation of Hg in different environmental media around the emission source, including soil/sediment, vegetables and grains [6–9]. Asia is the most important region of Hg emissions in the world during the past three decades [1,5,10–11], with China contributing to approximately 1/4 of global anthropogenic emission [5,12,13]. The Hg emissions from coal combustion, lead/zinc smelting and cement industry had become the main Hg contributing

sources in China in recent years [13,14].

Circulating fluidized bed (CFB) boiler and pulverized coal (PC) boiler are two major boilers of power plants. As a clean combustion technology, the CFB boiler has a wider fuel adaptability, higher combustion efficiency, lower pollutant emission and lower cost than that of PC boiler, resulting in extensive application in many cases [15]. China has the largest installed capacity of CFB boilers in the world, with more than 281 CFB boilers (installed capacity > 150 MW) in operation [16]. As for 2015, CFB boilers accounted for 10% of the total installed capacity in coal-fired power plants [17]. Hg released from coal combustion is an important source of anthropogenic Hg. Dai et al. [18] estimated the average Hg concentration in coal burned in China is $0.16 \text{ mg}\cdot\text{kg}^{-1}$ ($N = 1666$), and the world average value for Hg in coal is only $0.10 \text{ mg}\cdot\text{kg}^{-1}$ [19]. Coal combustion in power plants and industrial boilers is still the most dominant source of Hg emissions in

* Corresponding authors.

E-mail addresses: lizhonggen@mail.gyig.ac.cn (Z. Li), fengxinbin@mail.gyig.ac.cn (X. Feng).

<https://doi.org/10.1016/j.fuel.2019.04.071>

Received 21 September 2018; Received in revised form 29 January 2019; Accepted 11 April 2019

Available online 18 April 2019

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China [13,14,20].

Coal-fired power plants are among the top priority sources to be controlled in the Minamata Convention on Mercury. As the largest anthropogenic Hg emission source in China, it is particularly important to continuously update the quantity and emission factors of Hg emissions in coal-fired power plants [14]. Given the advantages of CFB boiler, its installations in China has been widely applied for new plants. However, Hg emission characteristics in CFB facilities have not been studied as extensively as those with PC boilers. Furthermore, technological progress of CFB boilers and air pollution control devices (APCDs) implementation also facilitate the need for updating of Hg emissions in this field. Most earlier studies measured total Hg (THg) content in flue gas or calculated the removal rate and emission factor of Hg [21–23]. There were few investigations focused on Hg emission speciation in the flue gas from CFB facilities [24]. Also earlier works showed diverse results because Hg concentration in coal had regional differences and the combination of APCDs were different in CFB power plants. Therefore, more onsite emission measurements of Hg from CFB boilers are urgent needed to reduce the uncertainties of anthropogenic Hg emission estimate from this source.

Guizhou has one of the largest coal reserves among the provinces in China and has incentives for coal-energy development [25]. In the past 20 years, coal-fired power plants have been greatly expanded in the province because of the west-to-east power transmission project. In addition, the CFB boilers were also applied in industrial production. In this study, the characteristics of atmospheric Hg emission, including emission quantity and speciation, at two CFB power plants in Guizhou that used different kind of coal or coal processing byproducts, along with different APCDs, were investigated. Based on the sampling and analyzing of flue gas using the Ontario Hydro Method (OHM), and analysis of Hg concentration in different input/output solid samples, Hg removal efficiency through APCDs and mercury atmospheric emission were calculated. The study updates mercury emission factor (MEF) of CFB boilers in China and provides directions for future study of this source.

2. Methodology

2.1. Power plants description and sampling sites

Guizhou Province, situated in southwest China, reserves large amounts of coal resources [25]. The studied two CFB power plants are located in the western and southwestern Guizhou for power plant #1 (P#1) and power plant #2 (P#2), respectively. The configuration of APCDs and the sampling locations of these two power plants were showed in Table 1 and Fig. 1. P#1 has two 150-MW CFB boilers equipped with in-furnace desulfurization (IFD) and electrostatic precipitator (ESP). P#2 has two 300-MW CFB boilers with selective non-catalytic reduction (SNCR), ESP, fabric filter (FF) and limestone gypsum wet flue gas desulfurization (WFGD). The fuel used in P#1 is anthracite coal and that of P#2 is gangue and coal slime from coal preparation plant, all fuels were from local area where the power plant located. Flue gas samples were collected at the ESP inlet and ESP outlet in P#1 and the ESP inlet and WFGD outlet in P#2, respectively (Fig. 1).

Table 1

Information on the studies CFB power plants and flue gas sampling locations.

Power plants	Boiler types	Installed capacity (MW)	APCDs ^a	Flue gas sampling locations
P#1	CFB	2 × 150 MW	IFD + ESP	ESP inlet and outlet
P#2	CFB	2 × 300 MW	SNCR + ESP-FF + WFGD	ESP-FF inlet and WFGD outlet

^a IFD: In-furnace desulphurization; ESP: Electrostatic precipitator; SNCR: Selective non-catalytic reduction; FF: Fabric filter; WFGD: Wet flue gas desulphurization.

2.2. Sampling and analysis methods

The Ontario Hydro Method (OHM) (ASTM Method 6784-02, 2008) was applied to sample gaseous elemental mercury (Hg^0), gaseous oxidized mercury (Hg^{2+}) and particulate mercury (Hg^p) in the flue gas by equipment of Model XC-572 (Apex Instruments, USA). Hg^p was first collected on the glass fiber filter, and an additional glass cyclone was used before the glass fiber filter when the flue gas contained too much particulate matter (PM) at locations such as ESP/FF inlet in this study. Three impingers each containing 100 mL KCl solution ($1 \text{ mol} \cdot \text{L}^{-1}$) were then used to capture Hg^{2+} . One impinge with $\text{H}_2\text{O}_2 + \text{HNO}_3$ (100 mL, 10% v/v + 5% v/v) and three impingers with $\text{H}_2\text{SO}_4 + \text{KMnO}_4$ solution (100 mL, 10% v/v + 4% m/v) jointly collected Hg^0 through oxidation and absorption processes. In the end, an impinger with silica gel was used to remove moisture in flue gas. Both the probe and the filters were heated to at least 120°C to avoid the water condensation and Hg adsorption on the tubing walls. Each flue gas sampling ran lasted for 1–1.5 h. Hg samples collected in the scavenging solutions were reduced by SnCl_2 , and measured 2-X times by Cold Vapor Atomic Absorption Spectrophotometry (CVAAS, F732S, Shanghai Huaguang Instrument Corp), which had an instrument detection limit of $0.05 \mu\text{g} \cdot \text{L}^{-1}$.

Solid materials including fuel (coal, gangue and coal slime), bottom ash, fly ash, limestone and gypsum were collected simultaneously during the flue gas sampling. After drying at 40°C for 3–7 days and grinding to $< 0.150 \text{ mm}$, Hg concentrations in solid samples, including the glass fiber filters, were determined using the US EPA Method 7473. The method heats solid samples at 800°C and measured the released Hg^0 by CVAAS (Lumex RA915+, Russia) with a method detection limit of $0.1 \mu\text{g} \cdot \text{kg}^{-1}$. Each solid sample was determined at least three times to obtain a mean Hg value. Proximate analysis of feed fuels was performed using the Chinese national standard method (GB/T 212-2008). Ultimate analysis (carbon, nitrogen and hydrogen) was determined by an elemental analyzer (vario MACRO cube, Elementar, Germany). Un-burned carbon (UBC) content in fly ash samples was also determined by the elemental analyzer. Total sulfur was detected by the Eschka method following GB/T 214-2007. The calorific value was determined using GB/T 213-2008.

2.3. Quality assurance and quality control

Materials used for flue gas sampling were carefully pretreated before each field work to reduce the system blank and possible contamination. Quartz filters for collecting Hg^p were heated for 2 h at 500°C . Teflon tubing was cleaned using 20% nitric acid (HNO_3), impingers and bottles were soaked into a 20% HNO_3 overnight, and then washed with deionized water before sampling. For Hg analysis, certified reference materials of coal (NIST SRM 1632d), fly ash (NIST SRM 1633c; GBW 08401), dolomite (JDo-1) and soil (GSS-5) were measured to guarantee the analytical quality. The difference between measured Hg and the recommended values was found to be less than 10%, such as the observed Hg concentrations in NIST SRM 1632d, NIST SRM 1633c, GBW08401, JDo-1, GSS-5 were 87 ± 8 ($N = 7$), 929 ± 137 ($N = 5$), 36 ± 2 ($N = 6$), 8.9 ± 1.5 ($N = 3$), 282 ± 27 ($N = 10$) $\mu\text{g} \cdot \text{kg}^{-1}$, respectively, versus to the recommended values: 92.8 ± 3.3 , 1005 ± 22 , 39, 9.5 and $290 \pm 30 \mu\text{g} \cdot \text{kg}^{-1}$, respectively. For proximate and ultimate analysis, the concentrations of certified reference materials (GSB06-2105-2007 for anthracite and GSB06-2182-2008-1

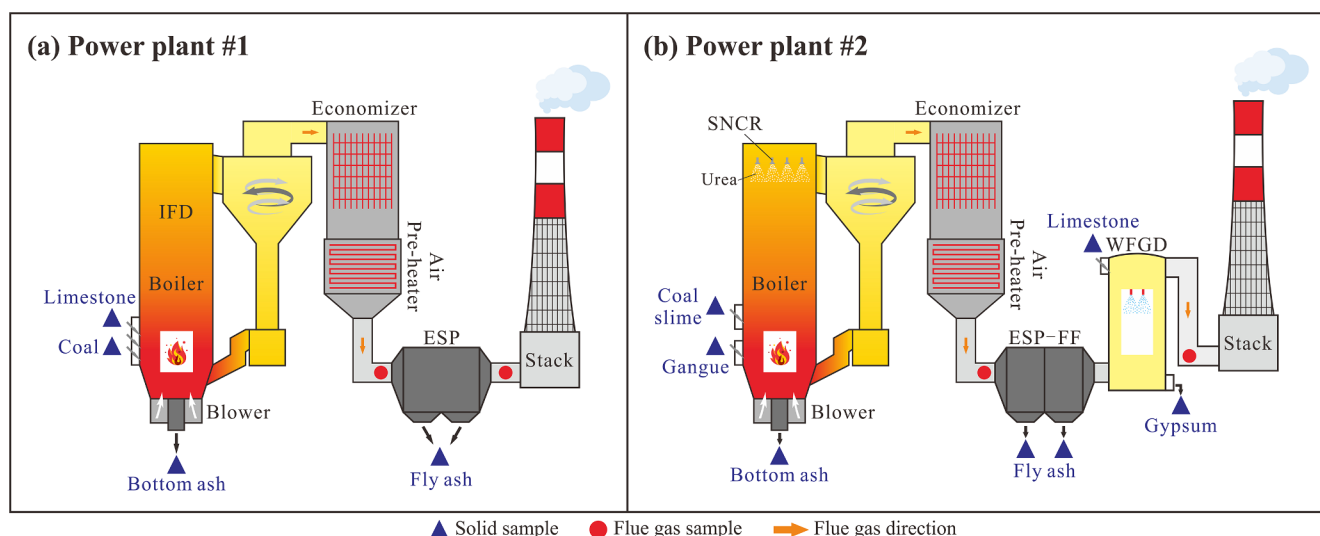


Fig. 1. Schematics of the two CFB power plants and the sampling locations.

for gangue) were measured with feed fuel samples (gangue, coal slime and coal). The means of measured compositions were within 5% of deviation from the certified contents.

3. Results and discussion

3.1. Fuel analysis

Proximate and ultimate analysis in fuels from the two plants are shown in Table 2. All feed coal samples were classified into middle to low calorific value coal according to GB 15224.3-2010 (16.7–21.3 MJ·kg⁻¹), since their calorific values were in the range of 17.03–19.66 MJ·kg⁻¹. The moisture content of coal slime (9.06%) was higher than that of other fuels (2.52–2.89%) and the volatiles of coal in P#1 (10.41%) was lower than that of P#2 (17.53–18.77%). Based on the National Coal Classification Standard of China (GB 15224.1-2010 and GB 15224.2-2010), the fuel in P#1 was classified as high-ash (39.62%) and high-sulfur (3.05%) coal, and the fuel in P#2 was extra-high-ash (43.94–45.15%) and low-sulfur (0.29–0.45%) coal. In summary, fuels in this study were of relatively high-ash and low-calorific values compared to typical coals used in PC boiler power plants [26,27], but were similar to feed coals used in six other CFB boilers in Shanxi province, China [21].

3.2. Mercury in solid samples

Hg concentration of coal in P#1 was $233 \pm 39 \mu\text{g}\cdot\text{kg}^{-1}$ (Table 3), exceeding the mean content in coal from China ($163 \mu\text{g}\cdot\text{kg}^{-1}$) [18] and the world average value of Hg in coal ($0.10 \text{ mg}\cdot\text{kg}^{-1}$) [19]. Gangue and coal slime in P#2 had Hg content of $113 \pm 23 \mu\text{g}\cdot\text{kg}^{-1}$ and $65 \pm 20 \mu\text{g}\cdot\text{kg}^{-1}$, respectively. The different Hg concentrations in these two plants may be related to the sulfur content since Hg in coal

were associated with sulfur in coal in most cases [28]. The limestone had low concentrations of Hg ($3\text{--}17 \mu\text{g}\cdot\text{kg}^{-1}$). In P#1, limestone was injected into the furnace and been calcined to CaO, which would interact with Hg in the flue gas to form the Hg^p and subsequently been removed by ESP as particulates. Limestone in P#2 was injected into the WFGD tower as limestone slurry to react with gaseous SO₂ to form gypsum. Hg in gypsum of P#2 was only $12 \pm 5 \mu\text{g}\cdot\text{kg}^{-1}$, this may be due to that most Hg in flue gas had been detained by ESP-FF in the upstream. Hg in fuels would be liberated into the flue gas under the combustion process in boilers and only few of Hg remained in the bottom ash [22]. Under the high combustion temperature (800–950 °C) and particulate circulation in CFB boilers, Hg content of bottom ash were 0 (below detection limit) $\mu\text{g}\cdot\text{kg}^{-1}$ and $2.8 \pm 2.4 \mu\text{g}\cdot\text{kg}^{-1}$ in P#1 and P#2, respectively. Hg in fly ash of P#1 was as high as $895 \pm 42 \mu\text{g}\cdot\text{kg}^{-1}$, while the figures for ESP fly ash and FF fly ash in P#2 were $237 \pm 6 \mu\text{g}\cdot\text{kg}^{-1}$ and $323 \pm 13 \mu\text{g}\cdot\text{kg}^{-1}$, respectively.

Mercury relative enrichment factor (MREF), a parameter that describes trace element enrichment in fly ash or bottom ash relative to the feed coal [29], is defined as:

$$MREF = \frac{C_{ash} \times A_{ad}}{C_{coal}} \quad (1)$$

where C_{ash} is Hg concentration in fly ash or bottom ash; A_{ad} is ash yield of feed coal on air-dried basis; C_{coal} is Hg concentration in feed coal.

Hg, a volatile element during coal combustion [29], was volatilized from feed coal and adsorbed onto and enriched in fly ash (MREF = 1.31–1.79, Table 3), while depleted in bottom ash (MREF = 0–0.02). It had been consistently observed that captured fly ash exceeded stack flue gas as the primary Hg output in recent studies for CFB utility boilers in China [21,30–33]. Factors influencing adsorption of Hg onto fly ash include particulate matter (PM) size, its structure and surface area [26], UBC content [34,35], flue gas

Table 2
Proximate and ultimate analysis of fuels.

Power plants	Fuel type	Proximate analysis (%)				Ultimate analysis (%)					Q _{net,ad} (MJ kg ⁻¹)
		M _{ad}	V _{ad}	A _{ad}	FC _{ad}	C _{ad}	H _{ad}	N _{ad}	S _{ad}		
P#1	Coal (No. = 5)	2.52	10.41	39.62	47.45	50.85	2.28	0.68	3.05	19.66	
P#2	Coal slime (No. = 4)	9.06	17.53	43.94	29.47	41.86	2.96	0.82	0.29	17.03	
	Gangue (No. = 4)	2.89	18.77	45.15	33.19	47.24	3.21	0.85	0.45	19.29	

Note: No., the number of samples; M, moisture; A, ash; V, volatile; FC, fixed carbon; C, carbon; H, hydrogen; N, nitrogen; O, oxygen; S, sulfur; Q_{net}, net calorific value; Subscript “ad”, air-dried basis.

Table 3
Hg concentration in solid samples.

P#1	Hg ($\mu\text{g}\cdot\text{kg}^{-1}$)	MREF [*]	P#2	Hg ($\mu\text{g}\cdot\text{kg}^{-1}$)	MREF [*]
Coal	233 \pm 39 (No. = 5)**		Coal slime	65 \pm 20 (No. = 4)	
Limestone	17 \pm 0 (No. = 1)		Gangue	113 \pm 23 (No. = 4)	
Bottom ash	0 \pm 0 (No. = 5)	0.00	Limestone	4 \pm 1 (No. = 4)	
Fly ash (ESP)	895 \pm 42 (No. = 5)	1.52	Bottom ash	2.8 \pm 2.4 (No. = 4)	0.02
			Fly ash (ESP)	237 \pm 6 (No. = 4)	1.31
			Fly ash (FF)	323 \pm 13 (No. = 4)	1.79
			Gypsum	12 \pm 5 (No. = 4)	

* MREF: mercury relative enrichment factor.

** Hg content is expressed in arithmetic mean \pm standard deviation.

compositions such as the halogen and sulfur [36,37], etc. Furthermore, the lower combustion temperature (800–950 °C) in CFB boiler than that in PC boiler (> 1400 °C) enhanced the activity of UBC that not been melted at the lower temperatures and possibly adsorbed more gaseous Hg (Hg^0 and Hg^{2+}) onto fly ash. The presence of CaO and high PM load in flue gas could further enhance the adsorption and facilitate in the Hg^p formation in CFB boilers. In the present study, the average UBC content in fly ash of P#1 and P#2 was 4.97% and 2.63%, respectively, resulting in Hg^p as the predominant species in the flue gas.

3.3. Mercury in flue gas and removals

Hg speciation and concentration in flue gas of the two investigated plants are shown in Table 4 and Fig. 2. In P#1, the flue gas temperature was maintained at 118–120 °C and total Hg (THg) in the flue gas dropped from $32.5 \pm 10.4 \mu\text{g}\cdot\text{m}^{-3}$ to $0.19 \pm 0.04 \mu\text{g}\cdot\text{m}^{-3}$ using the ESP. Impacted by the limestone slurry spray in WFGD, the temperature of flue gas decreased from 136 °C to 51 °C in P#2, and the THg level fell from $16.2 \pm 3.1 \mu\text{g}\cdot\text{m}^{-3}$ to $0.24 \pm 0.06 \mu\text{g}\cdot\text{m}^{-3}$ after ESP-FF + WFGD. THg concentration in the stack flue gas met the Hg emission standard for thermal power plants in China ($30 \mu\text{g}\cdot\text{m}^{-3}$, GB 13223-2011). The composition of Hg speciation in the flue gas of these two power plants was nearly the same (Table 4 and Fig. 2): both mainly comprised of Hg^p (95.11–99.26%), followed by Hg^{2+} (0.41–3.72%) and Hg^0 (0.33–1.17%) at the ESP inlet; while, Hg^{2+} (47.91–50.17%) and Hg^0 (44.94–51.15%) were the predominant species in the outlet of ESP/WFGD, consistent with an earlier report showing the predominance of Hg^{2+} and Hg^0 in Hg emissions from coal-fired power plants in China [31].

The removal efficiency of Hg by APCDs in the two CFB plants is shown in Table 5. A removal efficiency of 99.41% (ESP) and 98.53% (ESP-FF + WFGD) was achieved for THg in P#1 and P#2, respectively. Nearly all Hg^p was captured by APCDs and retained as fly ash, with the particulate Hg removal efficiency of 99.97% and 99.99% in P#1 and P#2 (Table 5), respectively. Although Hg^{2+} in the final discharged flue gas was both in a similar range of 0.10–0.12 $\mu\text{g}\cdot\text{m}^{-3}$, the removal efficiency of Hg^{2+} in P#2 (81.21%) was much higher than that in P#1 (24.10%), indicating that WFGD in P#2 had a great removal effect on Hg^{2+} [38]. Most of Hg in fuels was released into the flue gas as Hg^0 during the combustion, the addition of limestone in IFD significantly increased the particulate matter content in the flue gas [35], so gaseous Hg would be absorbed onto the particles as Hg^p after temperature dropping drastically [39]. At the same time, this also mean that Hg in flue gas of P#1 was easy to remove, and the removal efficiency can reach to the same level as P#2 (ESP-FF + WFGD) when P#1 only had ESP but with IFD. Therefore, desulphurization facilities (IFD or WFGD) in power plants can produce a substantial co-benefit in Hg removal in addition to SO_2 scrubbing.

Table 6 shows a summary of recent report of Hg emission through stack flue gas from coal-fired power plants. Compared with other studies, Hg concentrations of flue gas observed in this investigation was at a low level, similar to the values reported by Wei et al. [40] and Teng

et al. [23]. Due to the Hg removal efficiency of the PC boiler power plants (27.5–97.7%, most below 80%) was lower than that of CFB boilers (74.0–99.41%, most above 80%), the exhaust gas from CFB boilers had less Hg than that of PC boilers (Table 6). The combustion temperature of CFB and PC boilers can reach 800–950 °C and over 1400 °C, respectively [39]. Although both type of boilers released substantial fraction of Hg (> 90%) into flue gas during combustion [22], the CFB boilers was more conducive to remove the total Hg in flue gas since the majority Hg in flue gas exist as the Hg^p form which would be easily removed by ESP or FF [33,35,39]. This suggested that CFB boilers was in favor of Hg emission reduction when just equipped with dust collector.

3.4. Mercury mass balance and emission factors

A mass balance analysis was performed for the two power plants. The results are shown in Tables 7, 8 and Fig. 3. As the primary Hg input source for CFB boilers, fuels accounted 98.35% ($389.3 \text{ g}\cdot\text{d}^{-1}$) and 99.95% ($272.9 \text{ g}\cdot\text{d}^{-1}$) for THg input in P#1 and P#2, respectively. By comparison, the contribution from limestone was only $6.5 \text{ g}\cdot\text{d}^{-1}$ (1.65%) and $0.13 \text{ g}\cdot\text{d}^{-1}$ (0.05%), in P#1 and P#2, respectively. The temperature of combustion process in boilers was so high that little Hg remains in the bottom ash (< 0.56%). Hg in gypsum from WFGD desulphurization only accounted 0.24% of the THg output. However, most Hg^p , little Hg^{2+} and Hg^0 were captured in the fly ash by ESP or ESP-FF. Compared with other Hg output (such as stack flue gas and gypsum), fly ash was the leading discharge pathway for CFB utility boilers, this was similar to Duan et al. [33], Zhao et al. [26], Burmistrz et al. [41] and Teng et al. [23]. Therefore, the disposal of such Hg-containing solid wastes must be carefully managed to prevent Hg re-emissions into the environment. Based on the total Hg data obtained from the output (bottom ash, fly ash, gypsum and flue gas) and input (fuel coal and limestone), Hg output/input ratio ranged from 101.38–103.67% in the two studied plants (Tables 7, 8), indicating reliable calculation of Hg mass balance.

Mercury emission factor (MEF), a parameter estimating Hg emissions from coal-fired power plants, is calculated based on the following three equations:

$$MEF_1 = \frac{M_{\text{Hg}}}{M_{\text{coal}}} \quad (2)$$

$$MEF_2 = \frac{M_{\text{Hg}}}{P \times t} \quad (3)$$

$$MEF_3 = \frac{M_{\text{Hg}}}{M_{\text{coal}} \times Q_{\text{net,ad}}} \quad (4)$$

where MEF_1 , MEF_2 and MEF_3 are mercury emission factors based on the fuel consumption, electricity generated and heat value of fuel, respectively; M_{Hg} is the amount of Hg released into the atmosphere per day ($\text{g}\cdot\text{d}^{-1}$); M_{coal} is the consumption of fuel ($\text{t}\cdot\text{d}^{-1}$); P is the actual power generated by tested power plant (110 MW and 183 MW for P#1 and P#2, respectively); t is running time of generator ($24 \text{ h}\cdot\text{d}^{-1}$); $Q_{\text{net,ad}}$ is

Table 4
Total and speciated Hg in the flue gas and associated parameters of the flue gas in the two tested CFB power plants.

Power plants	Sampling location	Test number	Date	Time period	Sample Volume (m ³)	Temperature (°C)	Moisture content (%, v/v)	Hg ^P (µg m ⁻³)	Hg ²⁺ (µg m ⁻³)	Hg ⁰ (µg m ⁻³)	THg (µg m ⁻³)	Proportion of speciated Hg (%)			
												Hg ^P	Hg ²⁺		
P#1	ESP inlet	P#1-1	2014-08-26	12:50–14:02	0.94	117	4.93	25.0	0.13	0.11	25.2	99.05	0.51	0.44	
		P#1-2	2014-08-26	15:20–16:34	0.99	119	4.30	39.7	0.12	0.09	39.9	99.46	0.31	0.23	
	ESP outlet	Mean ± SD													
		P#1-1	2014-08-26	12:50–14:13	1.38	119	4.62 ± 0.44	32.3 ± 10.4	0.13 ± 0.00	0.10 ± 0.01	0.09	32.5 ± 10.4	99.26	0.41	0.33
		P#1-2	2014-08-26	15:20–16:44	1.32	120	5.72	0.00	0.08	0.08	0.16	2.80	50.88	46.32	
P#2	ESP-FF inlet	Mean ± SD			1.35 ± 0.04	120 ± 1	5.22 ± 0.70	0.01 ± 0.01	0.10 ± 0.02	0.09 ± 0.01	0.19 ± 0.04	4.89	50.17	44.94	
		P#2-1	2015-08-19	15:02–16:32	0.72	134	13.8	18.0	0.51	0.22	18.7	96.11	2.74	1.15	
		P#2-2	2015-08-19	16:53–18:08	0.63	132	13.8	15.6	0.88	0.15	16.6	93.79	5.30	0.91	
		P#2-3	2015-08-20	09:00–10:16	0.61	137	12.6	16.5	0.83	0.23	17.5	93.95	4.74	1.30	
		P#2-4	2015-08-20	10:45–12:01	0.64	140	15.0	11.4	0.25	0.15	11.8	96.58	2.10	1.31	
	WFGD outlet	Mean ± SD				0.65 ± 0.05	136 ± 4	13.8 ± 0.99	15.3 ± 2.84	0.62 ± 0.30	0.19 ± 0.04	16.2 ± 3.1	95.11	3.72	1.17
		P#2-1	2015-08-19	15:04–16:35	1.48	51	15.7	0.00	0.10	0.18	0.28	0.60	35.28	64.12	
			P#2-2	2015-08-19	16:55–18:10	1.17	51	14.8	0.00	0.17	0.11	0.28	1.19	60.30	38.51
			P#2-3	2015-08-20	08:43–10:13	1.40	50	15.0	0.00	0.14	0.10	0.24	0.47	57.98	41.55
		P#2-4	2015-08-20	10:29–12:00	1.59	50	12.8	0.00	0.06	0.09	0.15	1.51	38.08	60.41	
	Mean ± SD				1.41 ± 0.18	51 ± 1	14.6 ± 1.22	0.00 ± 0.00	0.12 ± 0.05	0.12 ± 0.04	0.24 ± 0.06	0.94	47.91	51.15	

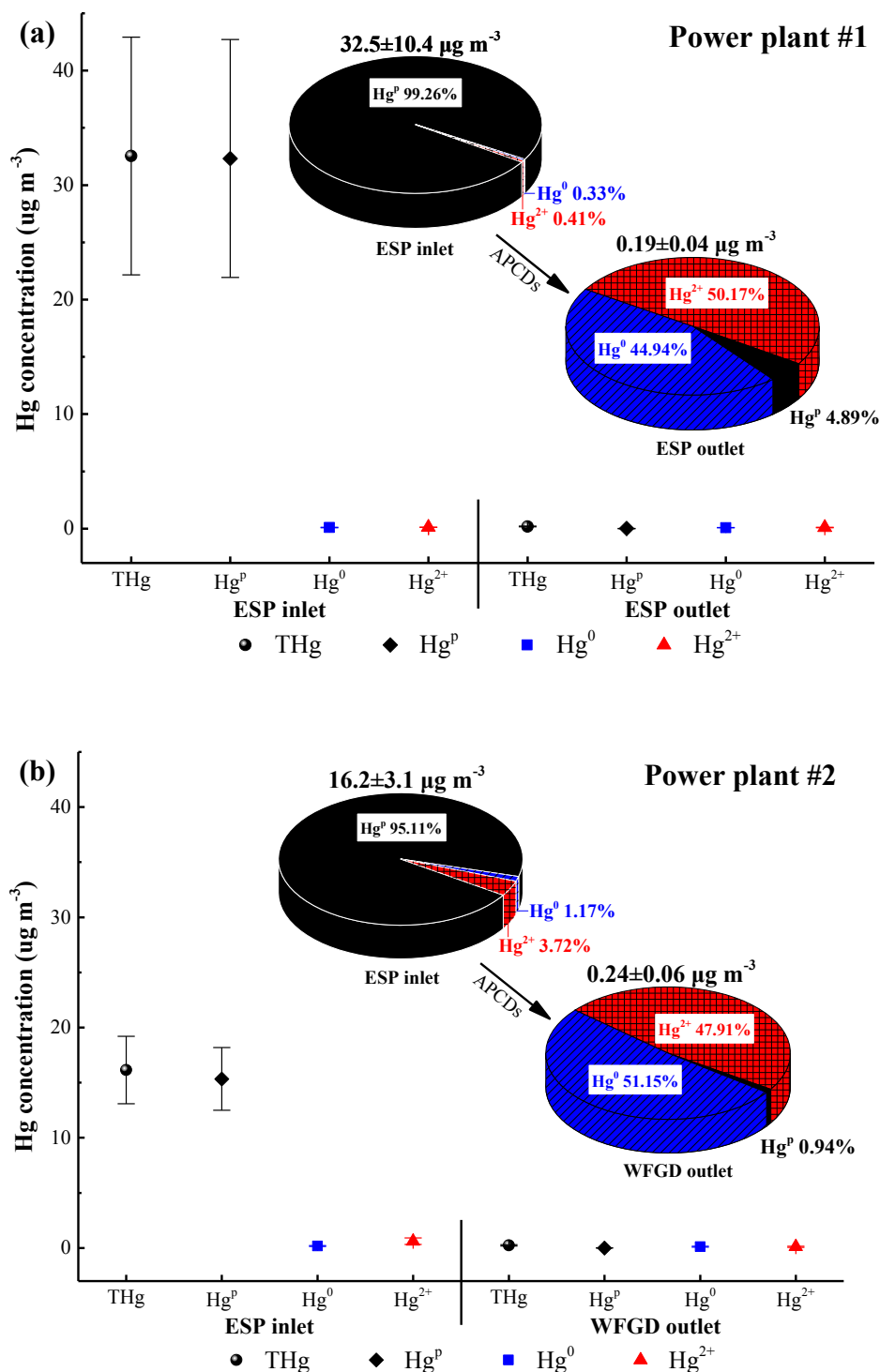


Fig. 2. Concentration and speciation of Hg in flue gas of two tested power plants.

Table 5
Removal efficiency of Hg in flue gas by the APCDs.

Power plants	APCDs	Hg ^p (%)	Hg ²⁺ (%)	Hg ⁰ (%)	THg (%)
P#1	ESP	99.97	24.10	15.19	99.41
P#2	ESP-FF + WFGD	99.99	81.21	36.46	98.53

calorific value of fuel ($\text{MJ}\cdot\text{kg}^{-1}$).

MEFs of tested CFB power plants are assessed in Tables 7 and 8. Compared the MEFs of coal-fired power plants reported in recent years

(Table 6), Hg release from the two CFB power plants ($1.19\text{--}1.54 \text{ mg Hg t}^{-1}$ coal; $0.78\text{--}1.19 \mu\text{g Hg}\cdot(\text{kWh})^{-1}$; $0.06\text{--}0.08 \text{ g Hg}\cdot\text{TJ}^{-1}$) was far less than others. The MEFs can be influenced by the fuel types, APCDs, boiler types, flue gas compositions, etc. [24,31,41]. With the technological development of APCDs and the elimination of small and outdated boilers, significant progress had been made in China in the past decade for air pollutants control, including NO_x , SO_2 and PM. Based on the data collected in this study, Hg release from CFB power plants was also substantially reduced as a result of modern APCD implementation, showing much lower Hg emission compared to earlier reports. Attention to Hg release from coal fired power plants needs to be directed to the Hg re-emission/release from the

Table 6
Comparison of Hg emission data from coal-fired power plants.

Geographical position	Boiler type	APCDs ^a	THg	Hg ^p	Hg ²⁺	Hg ⁰	Removal rate	Emission factors	Reference
Guizhou	CFB	IFD + ESP	0.19	0.01	0.1	0.09	99.41%	0.78 µg Hg·(kW·h) ⁻¹ 1.19 mg Hg·t ⁻¹ coal 0.06 g Hg·TJ ⁻¹	This study
Guizhou	CFB	SNCR + ESP-FF + WFGD	0.24	0	0.12	0.12	98.53%	1.19 µg Hg·(kW·h) ⁻¹ 1.54 mg Hg·t ⁻¹ coal 0.08 g Hg·TJ ⁻¹	This study
Shanxi	CFB	IFD + FF	–	–	–	–	85–90%	38.23 µg Hg·(kW·h) ⁻¹	[21]
Shanxi	CFB	SCR + FF + WFGD	–	–	–	–	89.42%	23.4 µg Hg·(kW·h) ⁻¹	[21]
Shanxi	CFB	IFD + ESP + FF	–	–	–	–	85–90%	18.11 µg Hg·(kW·h) ⁻¹	[21]
China	CFB	IFD + SNCR + ESP + WFGD	2.05	0.09	0.1	2.01	83.02%	0.69 g Hg·TJ ⁻¹	[24]
China	CFB	IFD + SNCR + FF + WFGD	2.26	0	0.04	2.22	76.41%	0.70 g Hg·TJ ⁻¹	[24]
China	CFB	IFD + SNCR + FF + WFGD	2.49	0	0.04	2.58	74.72%	0.80 g Hg·TJ ⁻¹	[24]
China	CFB	ESP	–	–	–	–	74%	–	[22]
China	CFB	FF	–	–	–	–	86%	–	[22]
China	CFB	SNCR + ESP-FF	0.27	–	–	–	90.30%	–	[23]
China	CFB	SNCR + FF + WFGD	0.3	–	–	–	83.80%	–	[23]
China	CFB	SNCR + FF + WFGD	0.25	–	–	–	84.40%	–	[23]
China	CFB	SCR + FF + WFGD	0.36	–	–	–	87.70%	–	[23]
China	CFB	FF + WFGD	0.26	–	–	–	99.35%	–	[40]
Spain	CFB	ESP	0.25	–	0.08	0.17	–	–	[42]
Spain	CFB	FF	2.15	–	0.39	1.76	–	–	[43]
China	PC	ESP + WFGD	0.37	–	–	–	97.65%	–	[40]
Anhui	PC	ESP + WFGD	–	–	–	–	36–46%	0.27 g Hg·t ⁻¹ coal	[27]
China	PC	SCR + ESP + WFGD	3.75	–	–	–	88.54%	–	[44]
Guizhou	PC	SCR + ESP + WFGD	0.6	0	0.11	0.49	97.56%	3.87 mg Hg·t ⁻¹ coal 2.54 µg Hg·(kW·h) ⁻¹	[45]
Guizhou	PC	ESP + WFGD	6.69	0	1.66	5.03	67.80%	75 mg Hg·t ⁻¹ coal	[30]
Guizhou	PC	ESP + WFGD	4.53	0	0.84	3.7	70.00%	43 mg Hg·t ⁻¹ coal	[30]
Guizhou	PC	ESP + WFGD	5.06	0	0.45	4.61	81.40%	32 mg Hg·t ⁻¹ coal	[30]
Guangdong	PC	ESP + WFGD	2.27	0	0.14	2.13	27.50%	25 mg Hg·t ⁻¹ coal	[30]
Beijing	PC	SCR + ESP + WFGD	1.22	0	0.13	1.08	35.50%	11 mg Hg·t ⁻¹ coal	[30]
Inner Mongolia	PC	SCR + FF + WFGD	0.57	0.01	0.03	0.53	96.20%	8.71 kg Hg·y ⁻¹	[45]
Inner Mongolia	PC	SCR + ESP + WFGD	3.8	0	0.42	3.39	77.74%	60.28 kg Hg·y ⁻¹	[45]
Inner Mongolia	PC	SCR + ESP-FF + WFGD	1	0	0.15	0.86	92.30%	8.95 kg Hg·y ⁻¹	[45]
China	PC	SCR + ESP + WFGD + WESP	4.3	0	0.3	4	56.59%	1.56 g Hg·TJ ⁻¹	[26]
Poland	PC	ESP + WFGD	2–5	–	–	–	65%	0.71–1.80 g Hg·TJ ⁻¹	[41]
Poland	PC	ESP + WFGD	15.1	–	–	–	60%	6.09 g Hg·TJ ⁻¹	[41]

^a IFD: In-furnace desulphurization; ESP: Electrostatic precipitator; SNCR: Selective non catalytic reduction; FF: Fabric filter; WFGD: Wet flue gas desulphurization; SCR: Selective catalytic reduction; WESP: Wet electrostatic precipitator; Unit of THg, Hg⁰, Hg²⁺ and Hg^p were in µg·m⁻³.

Table 7
Hg mass balance and emission factors of P#1.

Hg input/output	Materials	Material flow rate (t·d ⁻¹ for solid material/10 ⁴ × m ³ ·d ⁻¹ for flue gas)	Hg input/output (g·d ⁻¹)	Percentage (%)
Hg input	Coal	1730	389.30	98.35
	Limestone	384	6.54	1.65
	Total	/	395.84	100.00
Hg output	Bottom ash	240	0.00	0.00
	Fly ash	446	399.25	99.49
	Flue gas	1087	2.06	0.51
	Total	/	401.31	100.00
Hg output/input ratio		101.38%		
MEF based on fuel		1.19 mg Hg·t ⁻¹ coal		
MEF based on electricity [*]		0.78 µg Hg·(kW·h) ⁻¹		
MEF based on calorific value		0.06 g Hg·TJ ⁻¹		

* Actual output power: 110 MW.

captured fly ash.

4. Conclusions

Characterization of Hg emissions from CFB coal-fired power plants with different fuels and APCDs in Southwest China leads to the following conclusion:

Table 8
Hg mass balance and emission factors of P#2.

Hg input/output	Material	Material flow rate (t·d ⁻¹ for solid material/10 ⁴ × m ³ ·d ⁻¹ for flue gas)	Hg input/output (g·d ⁻¹)	Percentage (%)
Hg input	Gangue	1070	120.85	44.26
	Coal slime	2339	152.04	55.69
	Limestone	36	0.13	0.05
	Total	/	273.02	100.00
Hg output	Bottom ash	563	1.58	0.56
	Fly ash	1120	275.52	97.35
	Gypsum	58	0.69	0.24
	Flue gas	2213	5.24	1.85
	Total	/	283.03	100.00
Hg output/input ratio		103.67%		
MEF based on fuel		1.54 mg Hg·t ⁻¹ coal		
MEF based on electricity ^{**}		1.19 µg Hg·(kW·h) ⁻¹		
MEF based on calorific value		0.08 g Hg·TJ ⁻¹		

* Fly ash from ESP and FF account for 90% and 10%, respectively, and with weighted Hg content of 246 µg·kg⁻¹.

** Actual output power: 183 MW.

- (1) The stack flue gas from P#1 and P#2 had Hg concentrations measured by OHM of 0.19 µg·m⁻³ and 0.24 µg·m⁻³, respectively. Observed Hg speciation was: Hg⁰ (44.94–51.15%) and Hg²⁺ (47.91–50.17%), followed by Hg^p (0.94–4.89%).
- (2) The Hg removal efficiency of APCDs was in the range of

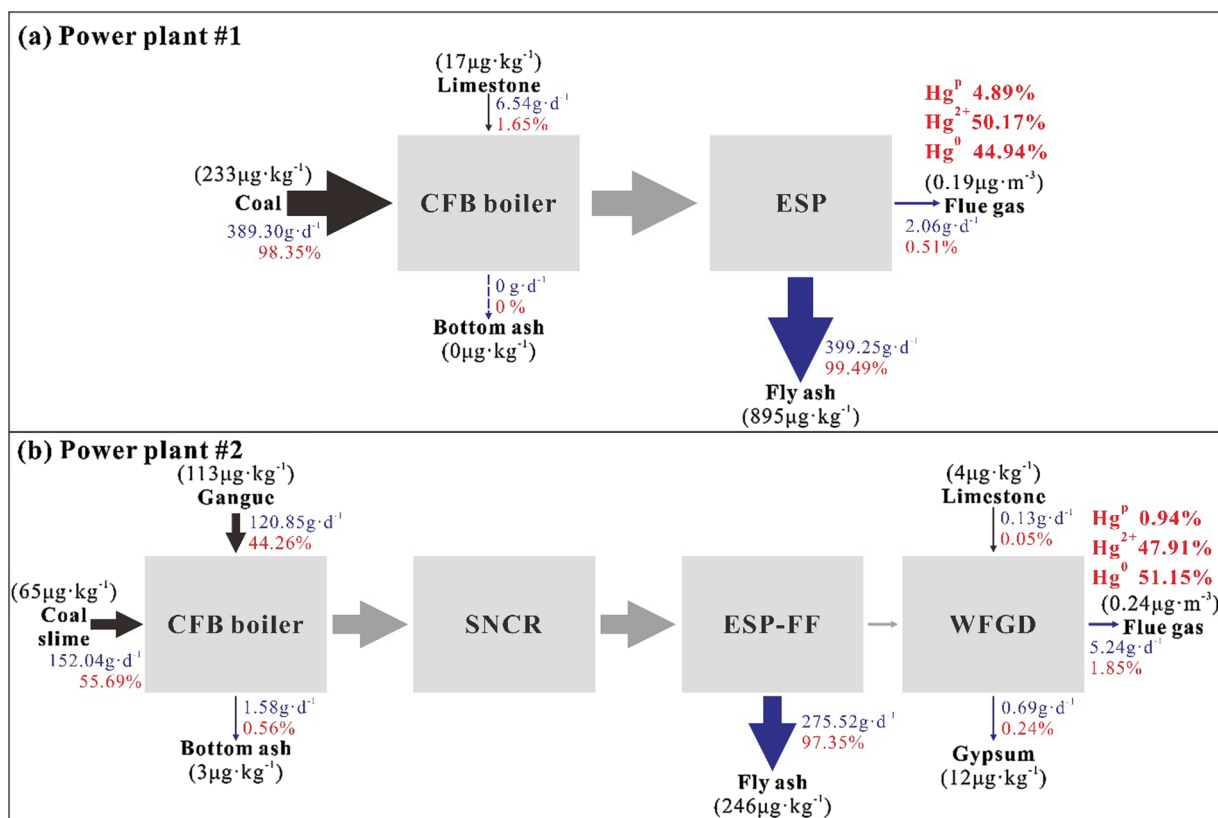


Fig. 3. Mass balance diagrams of Hg observed in the two CFB power plants.

98.53–99.41%, with most (> 97%) was removed by dust collector. As a consequence, MEFs were as low as 1.19–1.54 mg Hg t⁻¹ coal; 0.78–1.19 µg Hg·(kWh)⁻¹; 0.06–0.08 g Hg·TJ⁻¹.

- (3) The Hg output/input ratio ranged from 101.38 to 103.67%. Hg in the two CFB coal-fired plants came primarily from fuel (98.35–99.95%) and escaped mainly through the captured fly ash (97.35–99.49%). Besides, the Hg input from limestone and output from bottom ash and gypsum were in an extreme low level.
- (4) The MEFs estimated from the data collected in the two tested power plants were lower than the reports from earlier studies, suggesting that Hg emission data from CFB coal-fired power plants need to be further updated.

Acknowledgements

This work is financially supported by K.C.Wong Education Foundation, the National Key Basic Research Program of China (No. 2013CB430001), the Environmental Science and Technology Project of Guizhou Environmental Protection Department (No. Qian-Huan-Ke [2013]8) and the Natural Science Foundation of China (No. U1612442, 41428301). The authors greatly appreciate Zhuo Lu, Sheng Wang, Shengcai Liu and Changshun Yan for their hard work during the field sampling.

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