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Compressibilities of MnFe₂O₄ polymorphs

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Abstract The high-pressure behavior and stability of synthetic jacobsite MnFe₂O₄ have been investigated up to 39.55 GPa at room temperature by means of in situ synchrotron X-ray diffraction using diamond anvil cell and multi-anvil high-pressure apparatus. The MnFe₂O₄ spinel undergoes a phase transition at about 18 GPa to form a denser antiferromagnetic CaMn₂O₄-type (CM_{afm}) polymorph. The CM_{afm} MnFe₂O₄ is stable up to 39.55 GPa in this study and remains after decompression. Fitting the pressure-volume data using a third-order Birch-Murnaghan equation of state, the isothermal bulk modulus values and the first pressure derivatives were obtained as $K_0 = 169.7$ (35) GPa, $K'_0 = 2.87$ (40) for spinel-type $MnFe_2O_4$ and $K_0 = 149.2$ (24) GPa, $K'_0 = 3.98$ (19) for CM_{afm} MnFe₂O₄, respectively. If K'_0 is fixed to 4, K_0 was obtained as 160.6 (11) GPa for spinel-type MnFe₂O₄ and 148.9 (7) GPa for CM_{afm} MnFe₂O₄. The effects of cation substitution on the isothermal bulk modulus and pressure for phase transition of Fe³⁺-bearing spinels were discussed.

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Introduction

Spinel-structured minerals (spinels) are common accessory minerals in most kinds of crustal rocks. Spinels with the general chemical formula of AB_2O_4 can disorder over A and B sites, leading to a normal and an inverse structure. In the normal structure, divalent cations and trivalent cations are in tetrahedral site A and octahedral site B, respectively, while in the inverse structure, all divalent and half of the trivalent cations are in site B, with the rest of trivalent cations in site A (Levy et al. 2001). At ambient conditions, spinels have cubic structure (space group = $Fd\overline{3}m$, Z = 8).

The high-pressure phase transitions of spinels are of high relevance, because of the occurrence of their high-pressure polymorphs in nature (Chen et al. 2008) and their possible existence in the earth's mantle. Upon compression, phase transitions of cubic spinels to new denser polymorphs of CaFe₂O₄-type (space group Pnam) (e.g., MgAl₂O₄, Irifune et al. 1991; Ono et al. 2006), CaMn₂O₄-type (space group Pbcm) (e.g., Fe₃O₄, Fei et al. 1999), CaTi₂O₄-type (space group *Cmcm*) (e.g., ZnTi₂O₄, Wang et al. 2002) and ϵ -MgAl₂O₄-type (e.g., MgAl₂O₄, Liu 1978) have been reported. In the former three similar structures, Ca atoms are located in a dodecahedral site and Fe, Mn or Ti atoms in octahedral site. The octahedra share their corners and edges to form a compact network. The slight differences in modifications of the octahedra lead to two types of FeO₆ in CaFe₂O₄-type and a distorted MnO₆ in CaMn₂O₄type, whereas a more symmetric CaO₈ in CaTi₂O₄-type (Yamanaka et al. 2008). These almost indistinguishable differences lead to several debates on the properties and

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structure of post-spinel phases (Irifune et al. 2002). Additionally, a tetragonal distorted structure ($I4_1/amd$) could be adopted when ZnGa₂O₄ spinel was compressed to about 31 GPa (Errandonea et al. 2009) and MgCr₂O₄ spinel to about 20 GPa at room temperature (Yong et al. 2012). The tetragonal distorted structure of ZnGa₂O₄ changed to CaMn₂O₄type polymorph at 55 GPa (Errandonea et al. 2009).

Fe³⁺-bearing spinels are almost ubiquitous in lower crust and upper mantle, since the crystallization of magnetite (Fe³⁺-rich spinel) will cause the cessation of the crystallization of chromite (Cr^{3+} -rich spinel) (Roeder 1994). These spinels have been considered to contribute to the crustal magnetism (Frost and Shive 1986; Levy et al. 2000) and are important indicators of oxygen fugacity in host rocks (e.g., magnetite, O'Neill and Wall 1987). A phase transition of MgFe₂O₄ from spinel to CaMn₂O₄-type structure has been reported at 25 GPa (Andrault and Bolfan-Casanova 2001). This transition also happens for $ZnFe_2O_4$ at 24 GPa to either CaFe₂O₄- or CaTi₂O₄-like structure (Levy et al. 2000) and for Fe_3O_4 at 23.6 GPa (Fei et al. 1999). Besides, Wang et al. (2003) reported that $CoFe_2O_4$ adopted the CaFe₂O₄-like structure at high pressure. Among these highpressure experiments, a few spinels have been reported to decompose into oxide mixtures before post-spinel phase transition such as MgAl₂O₄ and MgFe₂O₄ (Ono et al. 2006; Levy et al. 2004). A comprehensive review on these transitions was published recently by Errandonea (2014).

In this study, we investigate the stability and compressibility of $MnFe_2O_4$ spinel up to pressure of 39.55 GPa at room temperature by means of in situ synchrotron X-ray diffraction using diamond anvil cell and multi-anvil highpressure apparatus. This ferrite spinel has an inverse spinel structure whose fraction of tetrahedral sites occupied by Mn^{2+} has been found to be 0.81 ± 0.03 at 300 K (Hastings and Corliss 1956). A phase transformation is observed for this compound, and the structure of new phase was refined. The elastic parameters of $MnFe_2O_4$ polymorphs were obtained by fitting the obtained pressure–volume data. Combined with previous results, the effects of cation substitution on the isothermal bulk modulus and pressure for phase transition of Fe³⁺-bearing spinels were discussed.

Experimental

The MnFe₂O₄ sample was prepared by a solid-state reaction. Reagent-grade MnO and Fe₂O₃ powders were mixed in the proportion corresponding to the MnFe₂O₄ stoichiometry, and the mixture was ground sufficiently and pressed into pellets with a diameter of 5 mm. The pellets were sintered at 1773 K for 24 h in argon atmosphere. A powder X-ray diffraction pattern confirms the sintered product is pure MnFe₂O₄ spinel. A mixture of MnFe₂O₄ plus 10 wt% Au, the internal pressure marker, was prepared for the highpressure in situ X-ray experiments.

Two kinds of high-pressure experiments were performed. The high-pressure angle dispersive in situ X-ray diffraction (ADXRD) experiment was carried out at beamline BL15U1 of SSRF, China, using a diamond anvil cell (DAC) with a culet of 300 µm. The experimental techniques used in this study were similar to previous studies (Wu et al. 2013; Xu et al. 2014). A rhenium disk with an initial thickness of 260 µm was pre-intended to a thickness of 32 μ m, where a hole with the diameter of 150 μ m was drilled as the sample chamber. The sample was loaded within the pressure medium of silicone oil into the chamber. A monochromatic X-ray with a wavelength of 0.6199 Å was adopted, and the beam spot on the sample was about 5 µm in diameter. Two-dimensional images were collected for 10 s at different pressures and integrated by Fit2D software to obtain the one-dimensional diffraction patterns (Hammersley et al. 1996). Pressure was calculated using the equation of state of Au proposed by Tsuchiya (2003) from the volume determined using (111) and (200) diffraction lines since other diffraction lines are not available, which leads the uncertainties of the pressure within ± 0.30 GPa, as shown in Table 1. XRD data were dealt with using Rietveld analysis by the GSAS-EXPGUI software (Larson and Von Dreele 2004) to obtain the lattice parameters.

On the other hand, the high-pressure energy dispersive in situ X-ray diffraction (EDXRD) experiment was conducted using a multi-anvil high-pressure apparatus, SPEED 1500, at beamline BL04B1 of SPring-8, Japan. The experimental method was similar to that described by Zhai et al. (2011, 2013). Kawai-type cell assembly composed of eight cubic second-stage tungsten carbide anvils with edge length of 26 mm and truncated edge length of 2 mm was adopted. A semi-sintered octahedron with 6.5 mm edge length made of MgO was used as the pressure medium, and pyrophyllite as the gasket. The tubing heater and sample capsule were made of TiB₂, for its high transparency for X-ray. A CCD camera and a Ge solidstate detector (SSD) were used to locate the sample and collect the X-ray diffraction patterns, respectively. During compression, the sample was heated to about 1000 K estimated from the temperature-power relationship of heater to release the deviatoric stress. The X-ray diffraction patterns were collected after quenching. The diffraction peak positions were determined using XrayAna program, and the lattice parameters were obtained by Refine program. Pressure was calculated using the equation of state of Au proposed by Tsuchiya (2003) from the volume determined using (111), (200), (220), (311) and (222) diffraction lines. Uncertainties in the pressure determination were mostly within ± 0.10 GPa, as shown in Table 1.

Table 1 Lattice parametersof spinel-type and CM_{afm} MnFe₂O₄ at various pressures

P (GPa)	Spinel-type			CM _{afm}			
	<i>a</i> (Å)	$V(\text{\AA}^3)$	P (GPa)	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	$V(Å^3)$
0.00 (7)	8.5157 (2)	617.54 (4)	0.00 (0)	9.8680 (1)	9.751 (1)	2.9435 (2)	566.46 (7)
0.83 (8)	8.4980 (4)	613.69 (9)	2.37 (6)	9.8293 (1)	9.704 (1)	2.9345 (2)	559.81 (6)
0.95 (5)	8.4979 (3)	613.67 (7)	3.74 (10)	9.7994 (1)	9.663 (1)	2.9234 (2)	553.64 (6)
1.75 (28)*	8.4830 (8)	610.51 (25)	5.39 (12)	9.7604 (1)	9.630(1)	2.9125 (2)	547.51 (6)
2.66 (7)	8.4718 (4)	608.03 (7)	7.14 (8)	9.7258 (1)	9.594 (1)	2.9023 (2)	541.62 (6)
2.89 (7)	8.4650 (5)	606.57 (11)	9.20 (12)	9.6915 (1)	9.548 (1)	2.8916 (2)	535.15 (6)
3.45 (20)*	8.4591 (9)	605.31 (26)	9.71 (17)	9.6829 (1)	9.542 (1)	2.8896 (2)	533.96 (6)
4.37 (15)*	8.4442 (8)	602.06 (26)	10.37 (14)	9.6652 (1)	9.531 (1)	2.8839 (2)	531.32 (6)
4.44 (18)	8.4436 (5)	601.99 (11)	11.40 (17)	9.6597 (1)	9.522 (1)	2.8797 (2)	529.75 (6)
5.47 (22)*	8.4220 (9)	597.29 (26)	13.31 (11)	9.6306 (1)	9.488 (1)	2.8715 (2)	524.77 (6)
5.79 (18)	8.4194 (4)	596.83 (8)	13.64 (8)	9.6171 (1)	9.472 (1)	2.8682 (2)	522.55 (6)
6.12 (13)*	8.4161 (9)	596.10 (30)	15.06 (11)	9.6108 (1)	9.465 (1)	2.8660 (2)	521.42 (6)
7.31 (12)	8.4045 (5)	593.60 (10)	17.36 (12)	9.5793 (1)	9.423 (1)	2.8532 (2)	515.11 (6)
7.43 (21)*	8.3960 (8)	591.94 (24)	18.90 (17)	9.5575 (1)	9.404 (1)	2.8490 (2)	512.14 (6)
10.34 (19)	8.3592 (4)	583.90 (8)	19.87 (12)	9.5451 (1)	9.393 (1)	2.8451 (2)	510.17 (6)
10.44 (30)*	8.3611 (9)	583.25 (23)	21.74 (14)	9.5196 (1)	9.363 (1)	2.8379 (2)	505.90 (6)
12.56 (29)*	8.3232 (8)	576.55 (29)	23.34 (11)	9.4880 (1)	9.336 (1)	2.8304 (2)	501.42 (6)
13.19 (21)*	8.3131 (9)	574.56 (23)	24.63 (16)	9.4747 (1)	9.320(1)	2.8258 (2)	499.05 (6)
13.49 (12)	8.3119 (4)	574.25 (8)	25.87 (19)*	9.4569 (1)	9.291 (1)	2.8170 (3)	495.00 (6)
16.87 (27)*	8.2721 (9)	565.98 (32)	26.79 (24)*	9.4389 (1)	9.279 (1)	2.8118 (3)	492.52 (6)
18.06 (9)	8.2536 (4)	562.25 (8)	28.82 (23)*	9.4234 (1)	9.249 (1)	2.8090 (3)	489.63 (5)
18.36 (23)*	8.2451 (9)	560.50 (22)	29.72 (21)*	9.4167 (1)	9.237 (1)	2.8056 (3)	488.02 (5)
			31.60 (14)*	9.3987 (1)	9.218 (1)	2.7978 (3)	484.73 (5)
			32.77 (16)*	9.3870 (1)	9.204 (1)	2.7949 (3)	482.90 (5)
			34.48 (12)*	9.3689 (1)	9.177 (1)	2.7906 (3)	479.80 (5)
			35.38 (13)*	9.3622 (1)	9.170 (1)	2.7896 (3)	478.93 (5)
			36.15 (14)*	9.3513 (1)	9.152 (1)	2.7846 (3)	476.57 (5)
			37.73 (16)*	9.3353 (1)	9.141 (1)	2.7806 (3)	474.48 (5)
			38.20 (27)*	9.3247 (1)	9.138 (1)	2.7795 (3)	473.66 (5)
			39.55 (23)*	9.3147 (1)	9.128 (1)	2.7729 (3)	471.48 (5)

Pressure values with * signals and corresponding lattice parameters are from DAC experiment. Number in parentheses represents the error of pressure or lattice parameter. The data for $CM_{\alpha\beta m}$ MnFe₂O₄ at ambient conditions were obtained after completely decompression

Results and discussion

High-pressure phase transformation and the new structure

Angle dispersive X-ray diffraction patterns were collected as a function of pressures up to 39.55 GPa at 300 K. Typical X-ray diffraction patterns are presented in Fig. 1a. A new weak peak appears at $2\theta = 13.604^{\circ}$ at 18.36 GPa and becomes stronger at 21.34 GPa, while the peaks of spineltype MnFe₂O₄ become weak and disappear after 22.31 GPa, indicating a mixture of the spinel and the new phase between 18.36 and 22.31 GPa due to the kinetics. It is noted that the peak broadening is observed above 10 GPa in Fig. 1a. This would be due to the deviatoric stress above 10 GPa when silicone oil was used as pressure medium (Klotz et al. 2009) and peak overlapping in high-pressure phase. The spinel and high-pressure phase coexisting over a rather large pressure range may also be owing to the pressure medium employed (Errandonea et al. 2005) and the slow kinetics at room temperature. The induced deviatoric stress by pressure medium and the peak overlapping yields a relatively larger error at higher pressure. The energy dispersive X-ray diffraction patterns were collected as a function of pressure up to 24.63 GPa, with typical ones presented in Fig. 1b. The X-ray diffraction pattern collected at 9.71 GPa after annealing clearly showed a phase transformation. The new phase did not transform back to spinel structure during



Fig. 1 Typical X-ray diffraction patterns of $MnFe_2O_4$ obtained by high-pressure ADXRD (a) and EDXRD (b) experiments. Abbreviations indexed to the diffraction peaks: Au = gold; * = X-ray fluorescence of Au

decompression after annealing as illustrated in Fig. 1b. The recovered sample of the new phase was checked by scanning electron microscope. In Fig. 2, there are only two phases, including the pressure maker Au (bright) and another homogeneous phase with the chemical composition of $MnFe_2O_4$ (gray). Therefore, $MnFe_2O_4$ spinel transforms to a high-pressure polymorph. The transforming pressure is about 10 GPa at about 1000 K and about 18 GPa at room



Fig. 2 Back-scattered electron image of recovered sample after annealing. The bright is Au and the gray is the high-pressure phase of $MnFe_2O_4$

temperature. Though the phase boundary between the spinel-type and the high-pressure phase of $MnFe_2O_4$ possibly has a negative dP/dT slope, which is similar to that of MgFe₂O₄ reported by Levy et al. (2004), further study is required to precisely determine the phase boundary since the kinetics equilibrium is difficult to reach at room temperature.

For the structure of the high-pressure $MnFe_2O_4$ phase, there are some candidates including $CaFe_2O_4$ -type, CaTi₂O₄-type, CaMn₂O₄-type and ε -MgAl₂O₄-type structures, as mentioned above. In order to refine the structure of the high-pressure phase, pure MnFe₂O₄ spinel was used as starting material and directly enclosed in TiB₂ heater to synthesize the high-pressure polymorph at about 12 GPa and 1473 K. An angle dispersive X-ray diffraction pattern of the quenched synthesized sample was additionally collected at ambient conditions at BL04B1 beamline, SPring-8, using monochromatic beam with energy of 51.00 keV. Because of the absence of the most intense lines, the ε -MgAl₂O₄-type structure was discarded. The Rietveld analysis was carried out based on CaFe₂O₄-type, CaTi₂O₄type and CaMn₂O₄-type models, and the method was similar to that described by Andrault and Bolfan-Casanova (2001). The results indicate that $CaFe_2O_4$ -type, $CaTi_2O_4$ type and $CaMn_2O_4$ -type models are comparable in fitting the experimental data with wRp and Rp of 0.1431 and 0.1001, 0.1322 and 0.0876, and 0.1413 and 0.0965, respectively. The Rietveld fitting results for the three structures are shown in Fig. 3. Based on the fitting results, it is quite difficult to determine the structure for the high-pressure $MnFe_2O_4$ polymorph. Therefore, other method is required to search the most suitable structure for the high-pressure



Fig. 3 Rietveld refinements of high-pressure $MnFe_2O_4$ polymorph's XRD pattern collected at ambient conditions. The *vertical bars* represent the high-pressure $MnFe_2O_4$ phase, and the *lower curve* represents the difference between observed and calculated profiles

 $MnFe_2O_4$ phase, such as theoretical simulations proved to be effective for determining the structural stability. Considering the magnetic ordering, we designed six candidates,



Fig. 4 Calculated relative enthalpy for ferromagnetic CaFe₂O₄- (CF_{*fm*}), CaTi₂O₄- (CT_{*fm*}) and CaMn₂O₄-type (CM_{*fm*}), and antiferromagnetic CaFe₂O₄- (CF_{*afm*}), CaTi₂O₄- (CT_{*afm*}) and CaMn₂O₄-type (CM_{*afm*}) MnFe₂O₄ as function of pressure

ferromagnetic CaFe₂O₄-type, CaTi₂O₄-type, CaMn₂O₄type and antiferromagnetic CaFe₂O₄-type, CaTi₂O₄-type, CaMn₂O₄-type (labeled CF_{fm}, CT_{fm}, CM_{fm}, CF_{afm}, CT_{afm}, CM_{afm}, respectively). The GGA+U method implemented in the VASP code (Kresse and Furthmüller 1996; Kresse and Joubert 1999) is employed to deal with the strong correlation effect of the Mn and Fe electrons in our sample. Based on a previous investigation on spinel MnFe₂O₄ (Huang and Cheng 2013), the values of U for the on-site Coulomb interaction in the localized d orbitals and J for the screened exchange energy are set as 4 and 0.70 eV for Mn, and 4.5 and 0.89 eV for Fe, respectively. All the selfconsistent calculations are converged until the total energy difference between electronic iterations being smaller than 10^{-7} eV per orthorhombic cell. The energy cutoff that determines the number of plane waves is 500 eV. The calculated different enthalpies compared to that of CF_{fm} are shown in Fig. 4. Obviously, CM_{afm} with the lowest enthalpy up to 40 GPa is proposed to be the most stable phase among those candidates. Thus, high-pressure MnFe₂O₄ polymorph in this work is assigned to be CM_{afm}, the antiferromagnetic CaMn₂O₄-type structure. The atomic positional parameters for the CM_{afm} MnFe₂O₄ were summarized in Table 2, and a structure model is illustrated in Fig. 5. The lattice parameters and volumes of CMafm MnFe2O4 at different pressures are refined, as listed in Table 1. The phase transition leads the coordination of Mn²⁺ from fourfold to eightfold, while Fe³⁺ remains, yielding a denser polymorph.

P-V equation of state

The third-order Birch–Murnagham (BM3) (Birch 1947) equation of state (EoS) was used to fit the high-pressure

Table 2 Refined unit cell and atomic positional parameters for the CM_{afm} MnFe₂O₄ at ambient conditions

Space group: Pmab; Z = 4 a = 9.8680 (1) Å, b = 9.751 (1) Å, c = 2.9435 (2) Å						
Atom	Wyck	Х	У	Z		
Mn	4d	0.75	0.1383 (10)	0.791 (5)		
Fe	8e	0.0757 (6)	0.1167 (8)	0.267 (4)		
01	4c	0.00	0.25	0.686 (18)		
O2	4d	0.25	0.205 (4)	0.169 (13)		
03	8e	0.117 (3)	-0.017 (4)	0.72 (4)		
Mn-O2	2.38 (4)	[1]	Fe-O1	2.28 (4)	[1]	
Mn-O2	1.89 (4)	[1]	Fe-O1	1.940 (35)	[1]	
Mn-O3	2.32 (8)	[2]	Fe-O2	1.946 (18)	[1]	
Mn-O3	2.28 (7)	[2]	Fe-O3	2.12 (9)	[1]	
			Fe-O3	1.91 (8)	[1]	
			Fe-O3	2.135 (35)	[1]	



Fig. 5 Schematic view of the high-pressure CM_{afm} MnFe₂O₄

P–V data, yielding the bulk modulus of cubic spinel- and orthorhombic CM_{afm} MnFe₂O₄. The BM3 EoS can be expressed as follows:

$$P = \frac{3K_0}{2} \left[\left(\frac{V_0}{V}\right)^{\frac{7}{3}} - \left(\frac{V_0}{V}\right)^{\frac{5}{3}} \right] \left\{ 1 + \frac{3}{4} \left(K'_0 - 4\right) \left[\left(\frac{V_0}{V}\right)^{\frac{2}{3}} - 1 \right] \right\}$$

where K_0 , K'_0 , and V_0 are the isothermal bulk modulus, its pressure derivation and the unit cell volume under ambient conditions, respectively. It is clear that silicone oil was used as pressure medium in the DAC experiment, and the non-hydrostatic effect is inevitable at higher pressures (Shen et al. 2004; Klotz et al. 2009). However, the X-ray diffraction peaks of sample and pressure marker Au were simultaneously collected in a very small spot. Therefore, non-hydrostatic effect is quite small, and all data from the DAC experiment can be used for fitting. The results from a least-squares fitting using an EosFit program (Angel 2001) are $V_0 = 617.1$ (2) Å³, $K_0 = 169.7$ (35) GPa, and $K'_0 = 2.87$ (40) for spinel-type MnFe₂O₄, and $V_0 = 566.6$ (2) Å³, $K_0 = 149.2$ (24) GPa, and $K'_0 = 3.98$ (19) for CM_{afm} MnFe₂O₄. The observed and refined unit cell volumes of spinel-type MnFe₂O₄ under ambient conditions are consistent with previous study (Passerine 1930). The unit cell volume data as a function of pressure and the compression curves calculated from the fitted parameters are plotted in Fig. 6. When the value of K'_0 was set as 4, the isothermal bulk modulus K_0 was obtained as 160.6 (11) and 148.9 (7) GPa for spinel-type and CM_{afm} MnFe₂O₄, respectively. In previous study, the adiabatic bulk modulus (K_{S0}) for spineltype MnFe₂O₄ was deduced as 161 GPa based on ultrasonic methods (Sakurai 1964). The orthorhombic CM_{afm} polymorph has a smaller isothermal bulk modulus compared with that of cubic spinel-type MnFe₂O₄. Due to the limitation of experimental technique, the degree of inversion has not been considered in this experiment. It may have little effect on the results since it could be covered by disturbances and uncertainties during measurement at high-pressure conditions (O'Neill and Dollase 1994).

In order to evaluate the quality of the Birch–Murnaghan EoS fitting, the relationships between the Eulerian strain $(f_{\rm E})$ and the normalized pressure $(F = P/[3f_{\rm E} (2f_{\rm E} + 1)^{5/2}])$ for both phases were plotted in Fig. 7, where $f_{\rm E}$ is defined as $[(V_0/V)^{2/3} - 1])/2$. The $f_{\rm E}$ -F plot provides a visual indication to whether higher-order terms, such as K'_0 and K''_0 (the isothermal bulk modulus' second-order pressure derivation



Fig. 6 Equation of state of the spinel-type and CM_{afm} MnFe₂O₄. Solid and open symbols are from multi-anvil and diamond anvil experiments, respectively. Solid curve represents the third-order Birch–Murnaghan equation fit for spinel-type phase with $K_0 = 169.7$ GPa and $K'_0 = 2.87$, and dashed curve represents the third-order Birch–Murnaghan equation fit for CM_{afm} phase with $K_0 = 149.2$ GPa and $K'_0 = 3.98$. The uncertainties of pressure and volume are within symbols



Fig. 7 Eulerian strain–normalized pressure (f_E -F) plots of spinel-type and CM_{*afm*} polymorphs of MnFe₂O₄. *Solid* and *open circles* represent the data of spinel-type and CM_{*afm*} phase, respectively. *Dotted lines* represent the linear fittings for both phases, respectively



Fig. 8 Variations of a/a_0 , b/b_0 , and c/c_0 for CM_{afm} MnFe₂O₄ as a function of pressure

at ambient conditions), are significant in the equation of state. The negative slope of the $f_{\rm E}$ -F plot indicates that the first pressure derivative of the bulk modulus is lower than 4. Therefore, the values of the derived K'_0 , 2.87 and 3.98 for spinel-type and CM_{afm} MnFe₂O₄, are consistent with the $f_{\rm E}$ -F plot analysis. The near horizontal slope of CM_{afm} phase is shallower than that of spinel-type phase, implying the derived K'_0 of high-pressure phase is larger than that of spinel phase. It is consistent with the fitted results.

The unit cell parameters of CM_{afm} MnFe₂O₄ as functions of pressure are plotted in Fig. 8. The divergence of the unit cell parameters indicates an anisotropic elasticity of CM_{afm} MnFe₂O₄. By fitting a "linearized" third-order Birch–Murnaghan EoS, simply by substituting the cube of the lattice parameter for the volume (Litasov et al. 2007; Zhai et al. 2011) and following the procedure implemented in the EosFit program (Angel 2001), we can obtain the axial compressible modulus parameters. For comparison, the K'_0 was fixed as 4. The analysis yields the axial compressibilities for *a*-, *b*- and *c*-axis are $K_a = 159.9$ (16) GPa, $K_b = 133.3$ (12) GPa and $K_c = 155.8$ (21) GPa, respectively, indicating the anisotropic elasticity along the axes.

The effect of A^{2+} cation substitution on the isothermal bulk modulus of Fe³⁺-bearing spinel and on the phase transition pressure can be evaluated by comparing the present results with previous studies, as summarized in Table 3. It is noted that different studies may yield various results for the same compound. The radii of Mg²⁺, Co²⁺, Zn²⁺, Fe^{2+} and Mn^{2+} with four coordination are 0.57, 0.58, 0.60, 0.63 and 0.66 Å (Shannon 1976), respectively. Based on the original data reported in references, the isothermal bulk modulus of different spinel was recalculated by fixing K'_0 as 4, as listed in Table 3. Generally, the isothermal bulk modulus K_0 of Fe³⁺-bearing spinel decreases with the increasing of radius of divalent cation, except that of Fe₃O₄. However, for the high-pressure polymorphs, CMafm MnFe2O4 shows a slight larger isothermal bulk modulus than those of CaMn₂O₄-type MgFe₂O₄ and CaFe₂O₄-type CoFe₂O₄, indicating that the isothermal bulk modulus of Fe³⁺-bearing post-spinel phase increases with the increasing of radius of divalent cation.

In previous studies, some Fe^{3+} -bearing spinels were reported to transform into orthorhombic high-pressure phases at ambient temperature or after annealing, as summarized in Table 3. The phase transition pressures of MgFe₂O₄, CoFe₂O₄, ZnFe₂O₄, and MnFe₂O₄ at ambient temperature are 30, 27, 24.5 and 18 GPa, respectively. After annealing, the phase transition of MgFe₂O₄, Fe₃O₄ and MnFe₂O₄ was observed at pressure of 25, 23.6 and 9.7 GPa, respectively, indicating that the phase transition pressure of the same Fe³⁺-bearing spinel decreases with heating. Therefore, it seems that the phase transition pressures of Fe³⁺-bearing spinels decrease with the increasing of radii of divalent cations.

Conclusions

High-pressure in situ synchrotron X-ray diffraction experiments of $MnFe_2O_4$ were carried out up to 39.55 GPa at room temperature by using diamond anvil cell and multianvil apparatus. A phase transition was observed at about 18 GPa, and the high-pressure phase was determined as a CM_{afm} structure. Fitting the *P*–*V* data to a third-order Birch–Murnaghan equation of state, the isothermal bulk modulus values and the first pressure derivatives were

Compound	Structure	K_0 (GPa)	K'_0	References	P_{Tr} (GPa)	References
MgFe ₂ O ₄	Spinel-type	195	4 (fixed)	Andrault and Bolfan-Casanova (2001)	25 ^b	Andrault and Bolfan-Casanova (2001)
	Spinel-type	209 ^a	4 (fixed)	Levy et al. (2004)	27.7	Wang et al. (2002)
	CaMn ₂ O ₄ -type	142	4 (fixed)	Andrault and Bolfan-Casanova (2001)	30	Greenberg et al. (2009)
CoFe ₂ O ₄	Spinel-type	178 ^a	4 (fixed)	Greenberg et al. (2009)	27	Greenberg et al. (2009)
	CaFe ₂ O ₄ -type	145	4 (fixed)	Wang et al. (2003)		
ZnFe ₂ O ₄	Spinel-type	176 ^a	4 (fixed)	Greenberg et al. (2009)	24.5	Levy et al. 2000
FeFe ₂ O ₄	Spinel-type	183	4 (fixed)	Mao et al. (1974)	23.6 ^b	Fei et al. (1999)
	Spinel-type	181	5.5	Nakagiri et al. (1986)		
	Spinel-type	215	7.5	Gerward and Staun (1995)		
	Spinel-type	217	4 (fixed)	Haavik et al. (2000)		
	Spinel-type	186	5.1	Reichmann and Jacobsen (2004)		
MnFe ₂ O ₄	Spinel-type	170	2.87	This study	18	This study
	Spinel-type	161	4 (fixed)	This study	9.7 ^b	This study
	CM _{afm}	149	3.98	This study		
	CM _{afm}	149	4 (fixed)	This study		

Table 3 Comparison of isothermal bulk moduli and pressure (P_{Tr}) of transformation for Fe³⁺-bearing oxide spinels and their high-pressure forms

^a Recalculated isothermal bulk moduli using data from the references

^b Result after annealing

obtained as $K_0 = 169.7$ (35) GPa, $K'_0 = 2.87$ (40) for spineltype MnFe₂O₄, and $K_0 = 149.2$ (24) GPa, $K'_0 = 3.98$ (19) for CM_{*afm*} MnFe₂O₄, respectively. If K'_0 is fixed to 4, K_0 was obtained as 160.6 (11) GPa for spinel-type MnFe₂O₄ and 148.9 (7) GPa for CM_{*afm*} MnFe₂O₄. Analysis of the axial compressibilities of orthorhombic CM_{*afm*} MnFe₂O₄ shows an anisotropic elastic behavior along the axes since the *b*-axis is more compressible than the *a*- and *c*-axis. Combined with previous results, the phase transition pressures of Fe³⁺-bearing spinels decrease with the increasing of radii of divalent cations.

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