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Pressure-dependent Raman spectra of β -Ca₃(PO₄)₂ whitlockite

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Abstract The pressure dependence of Raman spectra for whitlockite β -Ca₃(PO₄)₂ was investigated up to 18.0 GPa using a diamond-anvil cell at room temperature. The Raman frequencies of all observed bands for β -Ca₃(PO₄)₂ continuously increase with increasing pressure. The quantitative analysis of pressure dependence of Raman bands for the sample shows that the v_3 asymmetric and v_1 symmetric stretching vibrations are with the larger pressure coefficients (from 3.44 to 4.59 cm^{-1} GPa⁻¹) and that the v_4 bending and v_2 deforming vibrations are with the smaller pressure coefficients (from 1.46 to 3.12 cm⁻¹ GPa⁻¹). Combined with previous result, the isothermal mode Grüneisen parameters of β -Ca₃(PO₄)₂ were calculated. The splitting of the PO₄ symmetric stretching v_1 vibrations changes during compression and disappears around 15.4 GPa, which may be attributed to the evolution of PO_4 tetrahedra under high pressure.

Keywords Whitlockite $\cdot \beta$ -Ca₃(PO₄)₂ \cdot High pressure \cdot Raman spectra

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Introduction

The tricalcium phosphate [TCP, $Ca_3(PO_4)_2$] is one of the important phosphates. Many researchers, such as biologists, and inorganic and industrial chemists, are interested in this material since it is one of the most important biomaterials (Nurse et al. 1959; Rejda et al. 1977; Bigi et al. 1988; Elliott 1994) and it is with interesting luminescent properties (Sarver et al. 1961). TCP has four polymorphs of β -, α -, α' -, and γ -phase (Nurse et al. 1959; Murayama et al. 1986). The β -TCP is a resorbable phase and exhibits good biocompatibility (Jarcho et al. 1979; Kivrak and Tas 1998; Gibson et al. 2000), which has been widely used as a biocompatible material for bone replacement and for the coating of prosthetic implants as well as apatites (Elliott 1994). In fact, β -Ca₃(PO₄)₂ was initially labeled as whitlockite (Frondel 1941) and in rhombohedral structure (space group of R3c, Z = 21) (Gopal and Calvo 1972; Prewitt and Rothbard 1975; Dowty 1977; Yashima et al. 2003). Whitlockite was found as a common phosphate in lunar rocks and meteorites (Griffin et al. 1972; Delaney et al. 1984; Lundberg et al. 1988; Rojkovič et al. 1997; Anand et al. 2003; Ángeles et al. 2010; Xie et al. 2013).

Studies of the vibrational properties of β -TCP by means of Raman scattering measurement at ambient conditions have been reported (de Aza et al. 1997; Jillavenkatesa and Condrate 1998). The application of pressure usually has a large effect on the physical properties of materials. However, there is little work on the vibrational behavior of the β -TCP phase under compression.

In this paper, we report the first high-pressure micro-Raman spectroscopic study on the β -TCP up to 18.0 GPa using a diamond-anvil cell. The study has been performed over the frequency range from 300 to 1,200 cm⁻¹. The effect of pressure on the characteristic Raman active

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internal PO_4^{3-} modes of β -TCP is analyzed and discussed in relation to the crystal structure.

Experimental

High-purity β -TCP was prepared by a solid-state reaction from CaHPO₄ and CaCO₃. Reagent-grade CaHPO₄ and CaCO₃ powders were mixed in the proportion corresponding to Ca₃(PO₄)₂ stoichiometry, and the mixture was ground for 2 h in an agate mortar and pressed into pellets with a diameter of 5 mm under uniaxial pressure of 30 MPa. The pellets were sintered at 1,300 K for 48 h to form a single phase of β -TCP. The sintered product was crushed and ground into powder and characterized by powder X-ray diffractometer. The X-ray pattern confirmed the formation of a single β -TCP phase.

The high-pressure Raman spectroscopic measurements using a four-pin modified Merrill-Basset diamond-anvil cell were taken at Bayerisches Geoinstitut, Universität Bayreuth. The experimental method used in this study was similar to a previous study (Zhai et al. 2010). A Re metal plate with an initial thickness of 260 µm was used as gasket. The central area of the plate was pre-indented to a thickness of about 30 μ m, and a hole of 150 μ m in diameter was drilled through it. The synthetic β -TCP and a tiny ruby sphere were loaded with the Ne pressure medium into the hole in the gasket. The experimental pressure was determined using the ruby fluorescence method (Mao et al. 1978), and the estimated error is within 0.5 GPa in this study. Micro-Raman spectra were recorded in backscattering geometry using Dilor X-Y spectrometer with a liquid nitrogencooled CCD detector, $50 \times$ objective, and confocal mode. The digital resolution is about 1 cm⁻¹. Excitation was performed adopting an argon-ion laser with a wavelength of 514.5 nm and a power of 0.786 W. The acquisition time of each spectrum was 90 s. The Raman shift of each band was obtained by Lorentzian curve fitting to get a reasonable approximation.

Results and discussion

The β -TCP has a rhombohedral structure belonging to the *R3c* space group with *Z* = 21. In the this structure, phosphorus atoms are located at three crystallographic sites and tetrahedrally coordinated by oxygen atoms, and calcium atoms occupy five sites from Ca(1) to Ca(5). The coordinated numbers for Ca atoms from Ca(1) to Ca(5) are 7, 8, 8, 3, and 6, respectively (Yashima et al. 2003). The crystal structure of β -TCP can be described by the two kinds of columns of A and B types along the *c*-axis (Dickens et al. 1974; Elliott 1994; Yashima et al. 2003). The A column

is with the form of $-P(1)O_4-Ca(4)O_3-Ca(5)O_6-P(1)O_4-$, while the B column is of the form $-P(3)O_4-Ca(1)O_7-Ca(3)O_8-Ca(2)O_8-P(2)O_4-P(3)O_4-$. The B column is distorted and with a dense structure compared with the A column.

As described in previous study (de Aza et al. 1997), the factor group analysis of β-TCP yields 189 Raman active internal vibrations of PO_4^{3-} , including $7A_1 + 14E$ from the v_1 symmetric stretching mode, $14A_1 + 28E$ R from the v_2 deforming mode, $21A_1 + 42E$ from the v_3 asymmetric stretching mode, and $21A_1 + 42E$ from the v_4 bending mode. However, due to the low intensity and overlapping of a large number of internal vibrations, the number of observed Raman active bands is much less than 189. As shown in Fig. 1 and listed in Table 1, totally 13 Raman active internal vibrations of β-TCP were observed, including 4 vibrations of v_2 deforming mode, 4 vibrations of v_4 bending mode, 2 vibrations of v_1 symmetric stretching mode, and 3 vibrations of v_3 asymmetric stretching mode. The Raman spectrum was not analyzed by full range from 350 to 1,150 cm^{-1} , but by four separated ranges of 350-520, 520-660, 900-1,000, and 1,000-1,150 cm⁻¹ using PeakFit Program. Therefore, the peaks can be located well. Though the number of observed Raman peaks was less than that reported by de Aza et al. (1997), the typical characterization is similar.

It is noted that the intensity of the internal PO_4 symmetric stretching mode (v_1) located at 952 and 973 cm⁻¹ was strongest among the Raman peaks. An obvious splitting of the v_1 mode was observed, which is attributed to the effects of the distortions of the phosphate ions and the consequent spread of intratetrahedral bond lengths (de Aza et al. 1997). According to the high-resolution neutron powder diffraction results of β -TCP (Yashima et al. 2003), the P–O bond lengths of the PO₄ tetrahedra are from 1.510 to 1.574 Å with a variation of 0.064 Å. The averaged P-O bond lengths of $P(1)O_4$, $P(2)O_4$, and $P(3)O_4$ tetrahedra are 1.540, 1.538, and 1.535 Å, respectively. A longer bond length implies a weaker bond, i.e., smaller force constant and lower vibrational frequency according to Hooke's law. Therefore, the splitting of the v_1 symmetric stretching mode is due to the different types of nonequivalent PO₄ tetrahedra.

The Raman spectra of β -TCP were collected up to 18.0 GPa at room temperature. Figure 2 shows the typical Raman spectra of β -Ca₃(PO₄)₂ at different pressures. At high pressures, a broadband appears in the Raman shift region of 700–800 cm⁻¹, which is attributed to the diamond and becomes weak with increasing pressure. The Raman bands of β -TCP gradually shift to higher frequency with increasing pressure. This is reasonable since the P–O bond lengths become shorter with increasing pressures and shorter bond lengths imply stronger bonds, i.e., larger force constant, and consequently higher vibrational frequency according to Hooke's law. The Raman bands of the PO₄



Fig. 1 Raman spectrum of β -Ca₃(PO₄)₂ at ambient conditions

Table 1 Constants determined in the expression: $v_P = v_{i0} + \beta P$ and mode Grüneisen parameter (γ_{iT}) at room temperature for β -Ca₃(PO₄)₂

| Modes | ν_0 | v_{i0} | β | R^2 | γ_{iT} |
|-----------------------|---------|--------------|-----------|-------|---------------|
| <i>v</i> ₃ | 1,088 | 1,088.5 (4) | 4.59 (8) | 0.998 | 0.335 |
| | 1,036 | 1,038.8 (17) | 4.53 (24) | 0.980 | 0.347 |
| | 1,019 | 1,018.2 (6) | 3.50 (11) | 0.993 | 0.273 |
| ν_1 | 973 | 972.1 (9) | 3.44 (12) | 0.988 | 0.281 |
| | 952 | 952.9 (10) | 3.87 (13) | 0.989 | 0.323 |
| <i>v</i> ₄ | 630 | 630.3 (11) | 2.33 (15) | 0.965 | 0.294 |
| | 614 | 613.9 (11) | 1.90 (14) | 0.951 | 0.246 |
| | 590 | 594.3 (13) | 1.82 (17) | 0.923 | 0.243 |
| | 554 | 552.4 (8) | 1.46 (10) | 0.959 | 0.210 |
| <i>v</i> ₂ | 483 | 484.1 (11) | 3.12 (14) | 0.981 | 0.512 |
| | 461 | 458.8 (11) | 2.81 (15) | 0.976 | 0.487 |
| | 444 | 440.4 (12) | 2.33 (16) | 0.959 | 0.421 |
| | 411 | 410.0 (8) | 2.50 (11) | 0.983 | 0.485 |

 $v_{\rm P}$ and v_{i0} are in cm⁻¹ and β is in cm⁻¹ GPa⁻¹. v_0 was observed frequency (in cm⁻¹) at ambient conditions. R^2 is the correlation coefficient. Grüneisen parameter γ_{iT} was calculated with isothermal bulk modulus of $K_0 = 79.5$ GPa reported by Zhai and Wu (2010)

internal modes remain reasonably distinguishable before 15.4 GPa, where the splitting of the PO_4 symmetric stretching v_1 vibrations disappears and the two highest intensity bands merge into a single peak, as illustrated in Fig. 2. The pressure dependence of the PO_4 internal vibrational modes was determined using the data before 15.4 GPa.

Figure 3a shows the vertical enlarged Raman spectra between 380 and 680 cm⁻¹. Although the two peaks around 450 cm^{-1} are not easy to distinguish at ambient pressure, it is clear that four peaks are suitable for fitting in the range of 380–530 cm⁻¹, especially at high-pressure conditions. In the range of 1,000–1,150 cm⁻¹, nine peaks were reported



Fig. 2 Typical Raman spectra of β -Ca₃(PO₄)₂ at high pressures and room temperature. The band around 775 cm⁻¹ is ascribed to the diamond anvil

by de Aza et al. (1997). However, fewer peaks can be distinguished from the Raman spectrum at ambient pressure in this study. The vertical enlarged Raman spectra between 1,000 and 1,150 cm⁻¹ are shown in Fig. 3b. Here, three peaks were adopted to fit the spectra in the range of 1,000– 1,150 cm⁻¹. It is noted that below 10 GPa the three peaks can be clearly located. Therefore, for the v_3 asymmetric stretching mode, only data below 10 GPa were used for quantitative calculation.

Figure 4 presents the Raman shift versus pressure plot of β -TCP. It is obvious that the frequencies of phosphate modes for β -TCP vary linearly and continuously with pressure, implying that β -TCP is stable at room temperature through this pressure range. This is contrast with the behavior of β -TCP at high-temperature and high-pressure conditions (e.g., 950 °C and 4 GPa), where β -TCP transfers to γ -TCP (Roux et al. 1978; Murayama et al. 1986; Zhai et al. 2014). The pressure coefficients of PO₄ modes (Table 1) indicate that ν_3 and ν_1 bands in the higher-frequency region are more sensitive to pressure compared with the ν_4 and ν_2 bands in the lower-frequency region. In fact, the average pressure coefficients of ν_3 and ν_1 modes are 4.21 and 3.65 cm⁻¹ GPa⁻¹, respectively, whereas the



Fig. 3 Vertical enlarged Raman spectra of β -Ca₃(PO₄)₂ in the range of 380–530 cm⁻¹ (**a**) and 1,000–1,170 cm⁻¹ (**b**) at selected pressures and room temperature



Fig. 4 Pressure dependence of the Raman bands of $\beta\text{-}Ca_3(PO_4)_2$ at room temperature

average pressure coefficients for v_4 and v_2 are 1.88 and 2.69 cm⁻¹ GPa⁻¹, respectively.

The pressure coefficients of different Raman modes can be used to obtain the Grüneisen parameters which are extensively required in theoretical calculations. The Grüneisen parameter for each mode was calculated following the equation (Grüneisen 1912):

$$\gamma_{iT} = (\delta v_i / \delta P) K / v_i$$

where v_i is the frequency of the *i*-th mode, K is the bulk modulus. We used a zero pressure isothermal bulk modulus, K_0 of 79.5 GPa (Zhai and Wu 2010). The Grüneisen parameters are also listed in Table 1. The average value for the mode Grüneisen parameters of the bands associated to phosphate group vibrations is 0.343, which is comparable with the values of other phosphates, such as fluorapatite and tuite (Williams and Knittle 1996; Comodi et al. 2001a; Zhai et al. 2010). The bulk thermochemical Grüneisen parameter, which is equal to $\alpha KV/C_{\nu}$ (where α is the thermal expansion, K is the bulk modulus, V is the molar volume, and C_{v} is the volume constant heat capacity), is not available for β -TCP due to the lack of precise thermal expansion. However, for incompressible oxide compounds without polymerized tetrahedra, the bulk Grüneisen parameters are usually in the range from 0.8 to 2 (Shankland and Bass 1988). Actually, the evolution of the tetrahedral modes is not representative of the structural evolution on the whole. It is noted that the external modes including the vibrations associated to the calcium polyhedra can strongly



Fig. 5 The Raman shift difference (a) and relative intensity ratio (b) of splitting v_1 symmetric stretching vibration at different pressures and room temperature

affect the bulk Grüneisen parameter, because the behavior of a Ca–O polyhedron is much less rigid than that of the PO₄ tetrahedron under high pressure (Comodi et al. 2001a). Therefore, for β -TCP and other phosphates including tuite and fluorapatite, the relatively low average value of the PO₄ mode Grüneisen parameters indicates that the lattice modes and vibrations related to Ca²⁺ displacements must considerably largely contribute to the bulk thermochemical Grüneisen parameter than the PO₄ vibrations.

As mentioned above, the two highest PO_4 symmetric stretching v_1 vibrations approach and nearly merge into a single peak during compression. Indeed, at pressure of 15.4 GPa, it is difficult to resolve the two components of the v_1 symmetric stretching mode. However, it does not indicate a phase change combined with the result of X-ray diffraction measurement in previous study (Zhai and Wu 2010). The decreasing splitting of v_1 symmetric stretching mode in β -TCP is different from that of fluorapatite which shows decreasing splittings of v_3 asymmetric stretching and v_4 bending modes (Williams and Knittle 1996; Comodi et al. 2001a).

The Raman shift difference of splitting bands continuously decreases with increasing pressure, as shown in Fig. 5a. The reason is that the local crystal field surrounding the PO₄ groups deviates less from tetrahedral symmetry with increasing pressure. The intensity of high-frequency band (i.e., 973 cm⁻¹ at ambient conditions) decreases and low-frequency band (i.e., 952 cm⁻¹ at ambient conditions) increases during compression. The Raman intensity ratio of high-frequency band to low-frequency band $(R_{\rm b/l})$ at different pressure is plotted in Fig. 5b. It is obvious that the $R_{h/l}$ decreases continuously with compression. The intensity of Raman band was affected by polarizability derivatives of PO_4 . Therefore, the polarizability derivatives of PO_4 corresponding to both high-frequency and low-frequency vibrations of symmetric stretching v_1 mode change during compression, which might cause that the variations of the P-O bond lengths and bond angles of the PO₄ tetrahedra in the β -TCP structure become small and disappear at high pressure. There is no information about the evolution of PO_4 tetrahedra in the crystal structure of β-TCP under high pressure. Previous study shows that in fluorapatite, although the average P–O bond length decreases with increasing pressure, some P-O bond lengths decrease and then increase during compression (Comodi et al. 2001b). There is only one P atom in the crystal structure of fluorapatite, but three P atoms in β -TCP. Therefore, the variations of PO₄ tetrahedron in β -TCP are much more complicated than that in fluorapatite. It is surmisable that the changes of PO₄ tetrahedra in β -TCP are highly complex. Crystal structure of β -TCP is required under high pressure in the future.

Conclusions

The Raman spectra of synthetic β -Ca₃(PO₄)₂ up to 18.0 GPa at room temperature have been observed and analyzed in the region of 300–1,200 cm⁻¹. The Raman frequencies of all observed vibrations continuously and linearly increase with increasing pressure. The pressure coefficients of the PO₄ internal modes range from 1.46 to 4.59 cm⁻¹ GPa⁻¹ and the calculated isothermal mode Grüneisen parameters vary from 0.210 to 0.512, yielding an average mode Grüneisen parameter of 0.343. A splitting of the ν_1 symmetric stretching mode of PO₄ decreases with compression and disappears around 15.4 GPa, which probably is due to the change of PO₄ tetrahedra at high pressure.

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